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In-Depth Search Focused on Furans, Lactones, Volatile Phenols, and Acetals As Potential Age Markers of Madeira Wines by Comprehensive Two-Dimensional Gas Chromatography with Time-of-Flight Mass Spectrometry Combined with Solid Phase Microextraction

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ABSTRACT: The establishment of potential age markers of Madeira wine is of paramount significance as it may contribute to detect frauds and to ensure the authenticity of wine. Considering the chemical groups of furans, lactones, volatile phenols, and acetals, 103 volatile compounds were tentatively identified; among these, 71 have been reported for the first time in Madeira wines. The chemical groups that could be used as potential age markers were predominantly acetals, namely, diethoxymethane, 1,1-diethoxy-2-methyl-propane, 1-(1-ethoxyethoxy)-pentane, *trans*-dioxane and 2-propyl-1,3-dioxolane, and from the other chemical groups, 5-methylfurfural and *cis*-oak-lactone, independently of the variety and the type of wine. GC \times GC-ToFMS system offers a more useful approach to identify these compounds compared to previous studies using GC-qMS, due to the orthogonal systems, that reduce coelution, increase peak capacity and mass selectivity, contributing to the establishment of new potential Madeira wine age markers. Remarkable results were also obtained in terms of compound identification based on the organized structure of the peaks of structurally related compounds in the GC \times GC peak apex plots. This information represents a valuable approach for future studies, as the ordered-structure principle can considerably help the establishment of the composition of samples. This new approach provides data that can be extended to determine age markers of other types of wines.

KEYWORDS: Age markers, Madeira wine, HS-SPME, GC × GC-ToFMS

INTRODUCTION

Madeira wine is a fortified Portuguese wine produced in Madeira Island over the last centuries and plays an important role in the economy of the Island. The peculiar characteristics of Madeira wines arise from the specific and singular winemaking process. The fermentation process is stopped by the addition of natural grape spirit in order to obtain an ethanol content of 18-22% (v/v). Some wines undergo aging in wood casks in cellars at temperatures up to 30 °C, and humidity levels between 70 and 75%, while the majority of wines are submitted to a baking process, i.e., the wine is placed in large coated vats, and the temperature is slowly increased at about 5 °C per day and maintained at 45-50 °C during at least 3 months. After this treatment, the wine is allowed to undergo a maturation process in oak casks for a minimum of 3 years. Finally, some Madeira wines were submitted to an aging process, from a minimum of 3 to 20 years or even longer.^{1,2} The aging process in oak casks is fundamental for the Madeira wine's unique sensorial properties. During this period, several reactions and migration of molecules from the oak to wine can occur,^{3,4} which depends on some parameters, such as grape variety, wine making procedure, and oak characteristics (geographical origin, species of oak, seasoning of the staves, toasting, and age of cask),^{5–9} among others.

The establishment of potential age markers is important to detect frauds and to ensure the authenticity of the wine. Furthermore, the economic value of Madeira wine is highly associated with its age. Some volatile compounds that belong to furans, lactones, volatile phenols, and acetals have been reported as potential aging markers in Madeira wines.4,10-12 Compounds such as, 2-furfural, 5-methylfurfural, 5-hydroxymethylfurfural, cisoak-lactone, trans-oak-lactone, eugenol, guaiacol, m-cresol, ocresol, p-ethylphenol, maltol, vanillin, cis-dioxane, trans-dioxane, *cis*-dioxolane, and *trans*-dioxolane were considered.^{5,13-16} Furans (e.g., 2-furfural, 5-methylfurfural, and 5-hydroxymethyl-2-furfural) are formed by three pathways: pyrolysis of carbohydrates, dehydration of sugars through Maillard reaction, and caramelization,^{17–19} which occurs during winemaking and aging. As the levels of 2-furfural and 5-hydroxymethyl-2-furfural have a tendency to increase linearly during aging, they were considered as age markers.^{4,20} The lactones are important flavor compounds which are produced by cyclization of the corresponding hydro-xycarboxylic acids.^{4,21} Oak lactones, such as *cis*- and trans-oaklactone, are already present in natural oak, and their content increased due to seasoning and toasting,¹⁴ and from an organoleptic point of view, they are the most important lactones extractable from oak casks.²² Volatile phenols, like ethyl and vinylphenols, were also extracted from oak; nevertheless, their microbiological yeast transformation (e.g., Brettanomyces and

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Dekkara) from hydroxycinnamic acids of wine were reported as the main origin.²³ Acetals are formed during fermentation; nevertheless, their content increases significantly during the oxidative conditions of aging process. The high acetaldehyde content in wine contributes to the acetalization reaction with glycerol, which is favored at higher pH values, leading to four heterocyclic acetal alcohol formation: *cis*- and *trans*-5-hydroxy-2methyl-1,3-dioxane (*cis*-dioxane and *trans*-dioxane), and *cis*- and *trans*-4-hydroxymethyl-2-methyl-1,3-dioxalane (*cis*-dioxolane and *trans*-dioxolane). Heterocyclic acetal alcohols were identified and reported as potential age markers of Madeira wine.^{4,10,11} Other acetals, such as 1,1-diethoxyethane and 2,4,5-trimethyldioxolane, were also detected in table wines.²⁴

The Madeira wine volatile composition related to aging process has been studied using a one-dimensional chromatographic (¹D-GC) process, which revealed the complexity of this matrix.^{4,1 $\dot{0}$ -12} Although such a method often provides rewarding analytical results, in-depth analysis of the chromatograms frequently indicates that some peaks are the result of two or more coeluting compounds. Comprehensive two-dimensional gas chromatography (GC \times GC) was developed as a powerful separation method and emerged as an interesting alternative to analyze complex samples or analyze trace target analytes within a single analysis and overcoming the coelution problem.²⁵ The method employs two orthogonal mechanisms and is based on the application of two GC columns coated with different stationary phases, a nonpolar and a polar one (NP/P), sequentially linked through a modulator. Thus, the separation is ruled by boiling point properties in the first dimension (^{1}D) and polarity in the second one (^{2}D) .^{26,27} Therefore, two-dimensional gas chromatography (GC \times GC) offers faster running times, increased peak capacity, improved resolution and enhanced mass selectivity, good calibration linearity, and more sensitivity, and the limits of detection are improved due to the focusing of the peak in the modulator when compared to that in the one-dimensional GC. $^{28-30}$

In order to obtain a deeper characterization of the chemical groups potentially related with Madeira wine aging, namely, furans, lactones, volatile phenols, and acetals, the comprehensive two-dimensional gas chromatography with time-of-flight mass spectrometry ($GC \times GC-ToFMS$) combined with headspace solid-phase microextraction (HS-SPME) was used in the present research. This methodology was applied to Madeira wines from different varieties (Malvasia, Bual, Sercial, Verdelho, and Tinta Negra), types (sweet, medium sweet, dry, and medium dry), and ages (Vintage and blended wines). Finally, principal component analysis (PCA) was applied in order to establish potential age markers, which allow one to distinguish the different types of Madeira wines based on their age and even blends (average age).

MATERIAL AND METHODS

Samples. Twenty-three monovarietal Madeira wines from five *Vitis vinifera* L. grape varieties (one red, Tinta Negra, and four white, named as noble varieties of Madeira wine, Malvasia, Bual, Sercial, and Verdelho), aged from 3 to 20 years old (Y) and matured in oak casks, were used in this study. Tinta Negra is the main grape variety harvested in Madeira Island (Portugal) representing more than 80% of the vineyards. According to the age, the wines under study correspond to Vintage (a specific year of aged in casks, 17, 18, 19, and 20 years) and blended (B, an average aging period of 3, 5, 10, or 15 years) wines. Four types of wine were used: sweet (Malvasia, Tinta Negra), medium sweet (Bual, Tinta Negra), dry (Sercial, Tinta Negra), and medium dry (Verdelho, Tinta Negra), and

were aged in American oak casks (submitted to a lighter toasting). The ethanol content of the Madeira wines under study ranged from 18 to 19% (v/v). The samples were kindly provided by Madeira Wine Company, Madeira Island.

Reagents and Standards. Sodium chloride (99.5%, foodstuff grade) was purchased from Sigma Aldrich (Madrid, Spain), and ultra pure water was obtained from a Milli-Q system from Millipore (Milford, MA, USA). The retention index probes (*n*-alkanes series of C_8 to C_{20} straight-chain alkanes, concentration 40 mg/L in *n*-hexane) were supplied from Fluka (Buchs, Switzerland).

HS-SPME Methodology. The HS-SPME experimental parameters were previously established.² The SPME holder for manual sampling and fiber were purchased from Supelco (Aldrich, Bellefonte, PA, USA). The SPME device included a fused silica fiber coating partially cross-linked with 50/30 μ m divinylbenzene-carboxen-poly(dimethylsiloxane). Prior to use, the SPME fiber was conditioned at 270 °C for 60 min in the GC injector, according to the manufacturer's recommendations. Then, the fiber was daily conditioned for 10 min at 250 °C.

For the HS-SPME assay, aliquots of 1 mL of the sample were placed into a 5 mL glass vial. After the addition of 0.5 g of NaCl and stirring (0.5 \times 0.1 mm bar) at 400 rpm, the vial was capped with a PTFE septum and an aluminum cap (Chromacol, Hertfordshire, UK). The vial was placed in a thermostatted bath adjusted to 60.0 \pm 0.1 °C for 5 min, and then the SPME fiber was inserted in the headspace for 20 min. Each sample was analyzed in triplicate. Blanks, corresponding to the analysis of the coating fiber not submitted to any extraction procedure, were run between sets of three analyses.

 $GC \times GC$ -ToFMS Analysis. The $GC \times GC$ -ToFMS methodology was based on a previous study.² After the extraction/concentration step, the SPME coating fiber was manually introduced into the GC \times GC-ToFMS injection port at 250 °C and kept for 3 min for desorption. The injection port was lined with a 0.75 mm I.D. splitless glass linear. Splitless injections were used (30 s). LECO Pegasus 4D (LECO, St. Joseph, MI, USA) GC \times GC-ToFMS system consisted of an Agilent GC 7890A gas chromatograph, with a dual stage jet cryogenic modulator (licensed from Zoex) and a secondary oven. The detector was a highspeed ToF mass spectrometer. An HP-5 column (30 m \times 0.32 mm I.D., $0.25\,\mu m$ film thickness, J&W Scientific Inc., Folsom, CA, USA) was used as first-dimension column, and a DB-FFAP (0.79 m \times 0.25 mm I.D., $0.25\,\mu\mathrm{m}$ film thickness, J&W Scientific Inc., Folsom, CA, USA) was used as a second-dimension column. The carrier gas was helium at a constant flow rate of 2.50 mL/min. The primary oven temperature was programmed from 40 (1 min) to 230 °C (2 min) at 10 °C/min. The secondary oven temperature was programmed from 70 (1 min) to 250 °C (3 min) at 10 °C/min. The MS transfer line temperature was 250 °C, and the MS source temperature was 250 °C. The modulation time was 6 s; the modulator temperature was kept at 20 °C offset (above primary oven). A 6 s modulation time with a 30 °C secondary oven temperature offset was chosen to be a suitable compromise as it maintained the 1D separation, maximized the 2D resolution, and avoided the wrap-around effect (the elution time of a pulsed solute exceeds the modulation period) for compounds that were late to elute from the 2D. Ideally, all peaks must be detected before the subsequent reinjection, and hence, ${}^{2}t_{R}$ must be equal or less than the modulation period.^{31,32} The ToFMS was operated at a spectrum storage rate of 125 spectra/s. The mass spectrometer was operated in the EI mode at 70 eV using a range of m/z 33–350, and the voltage was –1695 V. Total ion chromatograms (TIC) were processed using the automated data processing software ChromaTOF (LECO) at a signal-to-noise threshold of 10. Contour plots were used to evaluate the separation general quality and for manual peak identification; a signal-to-noise threshold of 50 was used. Two commercial databases (Wiley 275 and US National Institute of Science and Technology (NIST) V. 2.0, Mainlib and Replib) were used. A mass spectral match factor, the majority (86%) of the tentatively



Figure 1. GC × GC total ion current chromatogram contour plot obtained from a sweet Madeira wine (Tinta Negra, 5 years); the chromatographic spaces corresponding to furans, lactones, volatile phenols, and acetals were highlighted. The *n*-alkanes series (C_6-C_{20}) was superimposed on the contour plot.

identified compounds showed similarity matches >850, was set to decide whether a peak was correctly identified or not. Furthermore, a manual inspection of the mass spectra was done, combined with the use of additional data, such as the retention index (RI) value, which was determined according to the Van den Dool and Kratz RI equation.³³ For the determination of the RI, a C_8-C_{20} *n*-alkanes series was used, and as some volatile compounds were eluted before C_8 , the solvent *n*-hexane was used as the C_6 standard. The RI values experimentally calculated were compared, when available, with values reported in the literature for similar chromatographic columns that employ as the first dimension the column of this study.^{34–51} The GC × GC area data were used as an approach to estimate the relative content of each volatile component.

Data Processing. In an initial approach, a linear regression was performed between total GC peak area of the chemical groups (furans, lactones, volatile phenols, and acetals) under study and wine age in order to establish potential age markers for Malvasia and Bual wines, and the results were expressed as r^2 (coefficient of determination). In a second step, PCA (principal component analysis) was applied to the autoscaled areas of the 103 volatile compounds tentatively identified by HS-SPME/GC × GC–ToFMS present in 23 monovarietal Madeira wines (from different varieties, types, and age) each with three replicates, using the R statistical software package.⁵² Autoscaling is a data pretreatment process that makes variables of different scales comparable. Each variable is autoscaled separately by subtracting its mean value and dividing by its standard deviation. The goal was to extract the main sources of variability and hence to help with the characterization of the data set.⁵³

RESULTS AND DISCUSSION

Contour and Peak Apex Plot Analysis. Automated processing of HS-SPME/GC × GC—ToFMS data was used to tentatively identify all peaks in the GC × GC chromatogram contour plots with a signal-to-noise threshold >50. The contour plot of the total ion chromatogram (Figure 1) exhibited several hundreds of peaks; however, this study was only focused on furans, lactones, volatile phenols, and acetals. The peak finding routine based on the deconvolution method allowed us to detect 103 compounds from these four chemical groups, which were tentatively identified on the basis of the comparison of their mass spectra to a reference database (MS) and by comparison of the RIs calculated (RI_{calc}) with the values reported in the literature (RI_{lit}) for the 5% phenylpolysilphenylene-siloxane (or equivalent) column (Tables 1 and 2). A range between 1 and 30 ($|RI_{calc}-RI_{lit}|$) was obtained for RI_{cal} compared to the RI_{lit} reported in the literature for one-dimensional GC with the 5%-phenyl-methylpolysiloxane GC column or equivalent. This difference in RI is considered reasonable (<5%) if one takes into account that (i) the literature data is obtained from a large range of GC stationary phases (several commercial GC columns are composed of 5% phenylpolysilphenylene-siloxane or equivalent stationary phases) and that (ii) the literature values were determined in a onedimensional chromatographic separation system, and the modulation causes some inaccuracy in the first dimension retention time.⁵⁴ In the case of the volatile compounds with $|RI_{calc}-RI_{lit}|$) values higher than 30, the information related to the mass spectra (m/z) was included in Tables 1 and 2.

Figure 1 shows the $GC \times GC$ total ion current chromatogram contour plot obtained from a sweet Madeira wine (Tinta Negra, 5Y); the chromatographic spaces corresponding to furans, lactones, volatile phenols, and acetals were highlighted. The nalkanes series $(C_6 - C_{20})$ used for the calculation of experimental RIs are also superimposed on the contour plot. The components of each chemical group were dispersed through the contour plot according to their volatility (¹D) and polarity (²D), and it becomes difficult to establish the two-dimension chromatographic space (GC \times GC) specific for each chemical group. As the principle of the structured chromatogram is very important in the identification, especially for the compounds that are not commercially available, a strategy was implemented to find this principle. Thus, peak apex plots were constructed, in order to find the possible structured 2D chromatographic profile, combining ${}^{1}t_{R}$ and ${}^{2}t_{R}$ values, for each chemical group under study, as shown for furans (Figure 2), lactones (Figure 3), volatile phenols (Figure 4), and acetals (Figure 5). Peak apex plots indicate the position of the maximum modulated peak of GC \times GC analysis, in the 2D chromatographic space.⁵⁵ For all chemical groups, as expected, it was observed that the decrease in volatility (high ${}^{1}t_{\rm R}$) is mainly related to the increase in the number of carbons.

The furans include several types of chemical structures; thus, they were organized in furan/alkyl furan, furanic aldehyde, furanic alcohol, benzofuran, furanic ester, and furanic acetal

Table 1	. Vol	latile (Compounds Identified by H	S-SPME/GC \times GC	-ToF	MS in	Dry (S	ercial a	id Tinta N	egra) and]	Medium Dr	y (Verde	lho and Ti	nta Negra)	Madeira V	Vines
									dry				mec	lium dry		
								Sercial		Tinta Neg	e.	>	erdelho		Tinta Negra	
							SY	3°]	0Y B 3	SY B S	Y B SY	B 1	0Y B 1	0Y ^f 33	ZB SY	В
Peak				Compounds previously identified	R.I. J	LI. R.I.										
number	$^{1}t_{R}^{a}(\mathbf{s})$	$^{2}t_{R}^{a}(s)$	Chemical groups	in Madeira wines	Calc ^b I	it. ^c Lit. ^d					peak area ^g (\times	10^5 (RSD ^h -	(%)			
			Furans													
1	84	0.46	Furan $(m/z = 68, 39)^i$		590 5	53 -	0.44	(23) 0.80	(13) -	- 0.69	(9) 0.63	(6) 0.71	(2) 1.05	(18) -	- 0.74	(2)
2	108	0.50	Tetrahydrofuran	ı	634 6	23 -		- 1.64	(3) 2.01	(27) 2.26	(16) 1.42	(36) 2.19	(11) 1.87	- (6)	•	,
8	192	1.80	3-Furfural		803 8	32 -	1.14	(9) 1.12	(6) 0.53	(2) 0.73	(9) 1.01	(3) 1.21	(1) 1.82	(12) 0.65	(12) 1.81	(21)
6	204	0.61	2-Ethoxytetrahydrofuran		812 8	25 -	16.30	(9) 18.6	6 (11) 8.19	(16) 15.4	(8) 11.81	(17) 15.7) (6) 25.58	(12) 11.53	(18) 20.10	(19)
10	216	1.90	2-Furfural	(4, 12, 59, 60)	837 8	29 838	480.36	(3) 565.0	6 (10) 338.	50 (16) 514.5	7 (2) 463.61	(8) 578.	75 (4) 575.0	9 (9) 443.71	l (14) 470.69	(4)
12	240	3.75	2-Furanmethanol	(12)	874 8	66 862	2.41	(13) 1.60	(11) 11.90	5 (4) 10.25	(10) 2.04	(4) 0.66	(14) 3.60	(11) 6.50	(7) 12.04	(20)
17	276	1.65	2-Acetylfuran	(4)	92.5 9	11 915	14.47	(18) 18.6	(3) 10.0 [°]	7 (7) 14.89	(6) 17.66	(2) 21.0	l (1) 42.07	(8) 12.20	(9) 18.69	(8)
29	324	1.68	5-Methyl-2-furfural	(11, 59, 60)	972 9	45 968	58.57	(8) 58.3	(3) 21.5	7 (10) 28.2	(6) 50.97	(3) 65.0) (4) 83.12	(7) 34.25	(12) 58.35	(10)
35	348	0.65	2-Pentylfuran		995 9	93 -	5.91	(12) 3.64	(1) 2.16	(21) 2.16	(18) 3.43	(7) 4.84	(5) 1.87	(9) 5.13	(17) 14.75	(11)
37	354	1.12	Ethyl 2-furoate	(4, 60)	1001 1	009 1000	9.28	(6) 12.70	(4) 3.59	(7) 5.30	(4) 8.02	(2) 10.2	7 (1) 16.26	(19) 4.71	(13) 6.39	(4)
38	354	1.22	Benzofuran		1001 1	005 1001	1.26	(5) 1.01	(1) 1.14	(4) 1.11	(2) 1.11	(9) 1.07	(4) 0.89	(11) 1.21	(10) 1.38	(3)
42	366	1.42	1-(2-Furyl)-1-propanone		1014 1	008 1011	2.66	(5) 2.41	(8) 1.06	(5) 1.08	(7) 2.56	(4) 2.52	(6) 5.45	(10) 1.73	(18) 3.01	(4)
45	390	1.45	2-Acetyl-5-methylfuran		1037 1	039 1039	2.71	(11) 3.74	(6) 2.76	(7) 1.87	(12) 2.83	(4) 3.61	(6) 4.63	(18) 2.29	(2) 3.50	(16)
51	408	1.44	S-Ethyl-2-furfural		1058 1	032 -	0.56	(7) 1.57	- (2)	•	- 0.81	(13) 0.91	(7) 0.69	- (2)	- 0.64	(9)
56	432	0.83	2-Diethoxymethylfuran		1082 1	- 870	4.23	(6) 2.57	(11) 1.28	(21) 2.41	(7) 3.55	(4) 1.59	(5) 4.64	(1) 1.39	(14) 2.87	(10)
57	432	3.94	5-Formylfurfural		1086 -		3.82	(9) 5.33	(4) 4.54	(12) 2.13	(16) 9.36	(12) 6.46	(13) 15.02	(16) 8.95	(19) 8.84	(10)
			$(m/z = 124, 123, 95)^i$													
60	438	1.48	3-Acetyl-2,5-dimethylfuran		1089 1	103 -	0.61	(6) 0.92	(12) 0.41	(18) 0.28	(19) 0.72	(8) 0.67	(10) 1.33	(17) 0.39	(10) 0.65	(17)
61	438	4.04	Methyl 2-furoate		1092 1	- 180	0.47	(13) 0.36	(14) 0.53	(10) 0.37	(10) 0.94	(11) 0.33	(4) 0.32	(16) 0.88	(8) 3.93	(17)
63	444	1.33	Furaneol		1095 1	- 060		- 1.15	(10) 0.27	(12) -	- 0.86	(9) 0.93	(1) 0.99	(10) 0.34	(20) -	,
65	456	1.09	2-Methylbenzofuran		1108 1	131 -	1.52	(13) 1.54	(7) 0.59	(8) 0.60	(20) 1.20	(10) 1.20	(9) 0.80	(17) 0.80	(9) 0.89	(19)
67	462	3.09	Maltol	(4)	1118 1	110 -	0.29	(6) 0.72	(19) 0.28	(18) 0.33	(4) 0.58	(10) 0.53	(4) 1.21	(5) 0.56	(19) 0.87	(2)
70	480	1.22	2-Methyl-5-propionylfuran		1136 1	151 -	0.19	(6) 0.21	(14) -	•	- 0.21	(18) 0.15	(16) 0.16	(16) -	- 0.18	(19)
78	564	1.69	5-Ethoxymethyl-2-furfural	(4)	1247 -		11.95	(3) 21.10) (7) 2.04	(13) 2.30	(14) 14.78	(8) 14.5	l (15) 28.48	(6) 5.94	(15) 8.29	(15)
			$(m/z = 126, 109, 81, 53)^i$													
79	588	2.64	S-Hydroxymethylfurfural	(4, 11, 12)	1286 1	256 -	0.99	(4) 3.27	(19) 3.04	(20) 6.36	(4) 2.97	(6) 6.08	(11) 5.52	(20) 5.76	(6) 11.06	(10)
85	624	2.92	S-Hydroxymethyldihydrofuran-		1328 -		34.08	(6) 64.3	(16) 21.4	4 (21) 28.48	(17) 31.12	(5) 30.7	5 (7) 41.78	(17) 17.46	(28) 32.08	(14)
			2-one													
			$(m/z = 85, 57, 29)^i$													
86	630	0.99	2-Methyl-2,3-dihydrobenzofuran	1	1332 1	385 -	69.0	(4) 0.67	(7) 0.83	(16) 0.74	(10) 0.90	(4) 0.59	(11) 0.80	(6) 0.74	(17) 1.32	(13)
			$(m/z = 134, 119, 91, 39)^{t}$													
			Subtotal				654.91	(2) 793.	07 (9) 438.	78 (13) 642.	54 (2) 635.10) (6) 772.	41 (3) 865.0	15 (5) 267.1	3 (11) 683.04	F (2)

Table 1. Continued

							dry				mediu	ım dry		
						Sercial		Tinta Negra		Vero	lelho		Tinta Negra	
					1	SY B ^e 10Y	B 3Y	B SY	B SY]	3 10Y	B 10	Y ^f 3Y.	B SY	В
Peak number	$^{1}t_{R}^{a}(\mathrm{s})$) $^{2}t_{R}^{a}(s)$	Chemical groups	Compounds previously identified in Madeira wines	R.I. R.I. R.I. Calc ^b Lit. ^c Lit. ^d			be	ak area g ($ imes$ 10	$^{5})$ (RSD ^h %)				
			Lactones											
13	240	4.90	Lactones 2,5-Furandione $(m/z = 98, 54, 26)^i$	1	881 828 - 0.6	8 (8) 1.97	(7) 0.44	(28) 0.73	(8) 0.17	(16) 0.47	(23) 5.26	(5) -		
20	282	2.53	γ -Butyrolactone	(4, 11, 12, 60)	932 915 920 57.	68 (19) 37.92	(1) 36.78	(8) 60.06	(13) 41.87	(16) 38.52	(13) 52.76	(12) 30.54	(14) 58.97	(19)
21 25	282 306	4.15 2.67	γ -Crotonolactone $(m/z = 98, 55,43)^i$		933 915 - 5.1 956 920 - 0.1	1 (11) 3.76 9 (8) 0.29	(3) 2.29 (1) 0.81	(15) 2.52 (19) 0.21	(15) 3.99 (5) 0.47	(16) 4.14 (12) 0.46	(10) 2.77 (11) 0.48	(8) 3.15 (13) 0.58	(12) 2.90 (7) 0.85	(-) (-)
26	306	2.99	3-Methylenedihydro-2,5-furandione	1	956 5.1	1 (32) 5.85	(29) 0.84	(21) 5.92	(14) 8.62	(7) 11.30	(8) 3.85	(11) 1.66	(12) -	
			$(m/z = 112, 84, 68, 40)^i$											
27	318	1.98	γ -Pentalactone		967 943 - 0.9	3 (13) 0.91	(3) 0.73	(10) 0.49	(18) 0.69	(9) 0.81	(10) 1.03	(16) 0.38	(11) 0.30	
34	342	2.55	α -Methyl- γ -crotonolactone		991 979 - 1.7	3 (12) 4.02	(8) 1.80	(20) 1.97	(12) 2.34	(4) 2.05	(5) 5.58	(16) 4.64	(10) 3.82	(5)
36	348	2.99	2H-Pyran-2-one		997 967 - 0.2	0 (28) 0.21	(5) 0.14	(3) 0.15	(17) 0.21	(27) 0.28	(12) 0.23	(6) 0.23	(13) 0.39	(8)
41	360	1.52	eta,eta-Dimethylbutylrolactone		1008 992 - 0.1	- (9) 6	•	•			•	- 0.25	(8) 0.39	(15)
4	384	4.38	3,4-Dihydro-3-methyl-2,5-furandione		1036 1057 - 1.7	3 (8) 4.98	(8) 0.66	(8) 4.76	(23) 4.23	(11) 4.30	(6) 6.14	(25) 2.15	(3) 2.12	(15)
47	396	1.66	Lavander lactone		1046 1039 1045 0.3	0 (3) 0.32	- (2)	•	•			- 0.32	(9) 0.30	(5)
48	396	2.02	3,4-Dimethyl-2,5-furandione		1046 1.2	9 (14) 1.16	(5) 0.56	(19) 0.67	(15) 0.99	(2) 1.18	(15) 1.12	(13) 0.63	(5) 4.29	(22)
ç	100		$(m/z = 126, 82, 54, 39)^i$									101 (0)	100 (01)	
49	390	06.5	Pantolactone $(m/z = 1.31, 71, 57, 43)$	(4, 12)	1050 - 1060 2.5	4 (15) 5.81	(9) I.33	(35) 4.34	(13) 4.19	10.5 (7)	(12) 3.93	(6) 1771	(17) 2.31	(0)
50	402	3.30	4-Methyl-2(SH)-furanone	1	1053 1036 - 0.3	7 (3) 0.79	(6) 0.35	(15) 0.28	(29) 0.34	(20) 0.35	(18) 0.34	(19) 0.50	(14) 1.71	(12)
52	408	1.76	γ -Hexalactone	(4)	1058 1055 1060 1.2	0 (14) 0.95	(14) 0.44	(10) 0.83	(13) 0.77	(11) 0.64	(11) 0.75	(17) 1.70	(26) 2.66	(11)
53	408	2.43	eta -methyl- γ -butyrolactone		1059	- 0.60	(11) 4.13	(22) 3.14	(6) 4.76	(14) 3.09	(34) 1.69	(8) 3.18	(18) 5.99	(14)
1	0	l	$(m/z = 85, 56, 41)^i$											
55	420	1.75	γ -Ethoxybutyrolactone		1071 1067 1069 5.8	9 (3) 20.33	(11) 4.36	(12) 13.41	(19) 11.15	(13) 1.43	(4) 11.53	(15) 3.64	(8) 5.86	(2)
71	480	3.87	Solerone $(m/z = 118, 56, 41)^{t}$	(4)	1140 1107 - 0.3	8 (8) 1.15	(8) 0.34	(8) 0.32	6 0.76	(9) 0.88	(2) 0.90	(16) 0.38	(8) 0.43	(7)
72	492	2.43	γ -Heptalactone		1152 1130 1162 0.4	4 (18) 0.81	- (6)	•	- 0.84	(16) 0.47	(8) 0.78	- (9)		
73	510	2.86	Mevalonic lactone		1174 1156 - 1.2	2 (4) 2.29	(6) 0.40	(20) 1.18	(14) 1.84	(15) 0.95	(10) 1.88	(11) 0.38	(16) 1.04	(5)
80	588	1.50	γ -Octalactone	(4, 60)	1284 1287 1291 0.6	3 (6) 0.75	(7) 0.44	(10) 0.46	(5) 0.80	(18) 0.55	(8) 0.59	(9) 0.67	(11) 1.30	(13)
83	618	1.26	trans-oak-lactone	(4, 11, 60)	1320 1292 1297 7.7	2 (2) 14.80	(10) 0.50	(12) 1.33	(9) 7.81	(8) 9.17	(10) 24.72	(16) 1.62	(19) 2.02	(15)
88	642	1.37	cis-oak-lactone	(4, 11, 60)	1345 1340 1329 24	84 (6) 58.99	(7) 2.37	(8) 4.85	(10) 28.14	(4) 32.03	(9) 52.87	(22) 4.94	(11) 6.54	(13)
89	660	3.39	2-Benzofuran-1(3H)-one		1366 0.2	9 (15) 0.41	(19) -	- 0.13	(19) 0.36	(8) 0.38	(15) 0.31	(25) 0.29	(15) 0.63	(16)
			$(m/z = 134, 105, 77, 51)^{i}$											
91	672	1.44	γ -Nonalactone	(4, 60)	1377 1366 1368 2.2	5 (11) 2.61	(12) 1.66	(18) 1.15	(11) 3.83	(6) 2.13	(5) 1.39	(10) 2.30	(17) 3.98	(2)
96	750	1.40	γ -Decalactone	(4, 60)	1471 1472 1472 0.6	9 (6) 0.61	(8) 0.64	(18) -	- 0.87	(4) 0.47	(18) 0.24	(13) 0.74	(13) 1.30	(17)
67	762	1.64	Massoia lactone		1487 1508 1482 0.3	7 (9) 0.47	(11) 0.12	(27) 0.10	(26) 0.33	(9) 0.23	(6) 0.15	(12) 0.25	(17) 0.33	(18)
100	900	1.25	γ -Dodecalactone		1684 1685 - 0.4	3 (19) 0.27	(9) 0.21	(16) 0.09	(1) 0.57	(18) 0.13	(7) 0.07	(13) 0.20	(15) 1.24	(11)
101	918	1.33	δ -Dodecalactone		1713 1721 - 0.3	0 (23) 0.20	(22) 0.15	(15) -	- 0.14	(24) 0.17	(11) -	- 0.28	(6) 0.84	(10)
103	966	0.87	Muskolactone		1841 1839 - 2.5	4 (12) 1.15	- (6)	•	- 3.21	(11) 1.20	(12) -	•	- 4.32	(8)
			Subtotal		12	7.22 (8) 174.39	(5) 62.49	(4) 109.10	(11) 134.27	(6) 120.81	(3) 181.36	(9) 66.83	(8) 116.81	(10)

Table 1. Continued

		в			(33) (24)	(18)		(13)	(5)	(19)	(5)	(13)	(3)	(17)	(6)	(34)	(2)	(2)	(7)	(8)	(7)			(8)	(12)	(6)	(13)	(9)	(15)	(10)	(21)	(2)			(16)
	Tinta Negra	SY			(17) 10.270.33	(14) 2.99	(6) 1.88	(3) 3.11	(8) 0.41	(4) 2.27	(13) 0.65	(11) 3.79	(12) 1.30	0.26	(15) 2.41	(11) 0.26	(10) 3.40	(16) 0.74	(21) 0.71	(13) 2.70	(7) 37.50		,	(5) 16.29	(6) 82.70	(19) 242.88	(8) 8.32	(4) 14.62	(6) 34.23	20) 0.71	(11) 9.53	(26) 8.23			(25) 4.59
ry		3Y B) 3.78 (1.79 () 0.82 (1.91 (0.20) 0.25 (0.44 () 0.77 (1.22 () 1.02 (0.09 (0.67 () 0.50 (0.61 () 0.76 (14.82 (18.62 (71.32 (284.80 (12.21 (9.11	15.36 () 0.54 () 4.21 () 5.88 (4.43 (
medium di		$10Y^{f}$			98 (13 61 (8)	(9) 96	67 (14	12 (8)	57 (9)	28 (11	43 (6)	27 (10	77 (7)	ı	92 (17	22 (4)	40 (9)	59 (10	33 (7)	31 (19	6.41 (8)		,	5.78 (8)	6) (6)	56.11 (2)	1.43 (7)	5.17 (7)	(8) (8)	85 (20	0.22 (18	11 (22			91 (5)
1	rdelho	ΥB	\sim		(9) (4)	(10) 0.	(18) 0.	(7) 2.	(10) 0.	(7) 7.	(7) 0.	(8) 0.	(9) 1.	(12) -	(18) 0.	(18) 0.3	(10) 4.	(12) 0.	(17) 0.	(12) 0.	(6) 26		•	(9) 55	5 (19) 55	4 (19) 36	(4) 31	(6) 25	(19) 41	(14) 0.	(5) 10	(9) 8.			(11) 3.
	Ve	B 10	0 ⁵) (RSD ^h %		(4) 8.48 (13) 0.25	(11) 1.44	(11) 0.52	(4) 1.46	(5) 0.62	(11) 7.39	(6) 0.31	(15) 0.46	(14) 7.77	(12) 0.35	(10) 0.84	(20) 0.23	(14) 1.32	(12) 1.11	(13) 0.23	(15) 0.49	(8) 33.28			(13) 38.65	(12) 137.35	(18) 271.14	(10) 4.97	(2) 38.22	(9) 45.33	(14) 0.29	(15) 7.09	(5) 3.45			(14) 2.09
		SY	t area ^g ($ imes$ 10		(8) 1.370.27	(2) 1.93	(16) 1.29	(10) 2.27	0.75	(12) 4.39	(13) 0.58	(3) 1.96	(5) 3.47	(2) 0.30	(18) 1.19	0.17	(10) 1.43	1.05	(4) 0.35	(5) 1.07	(4) 23.85		(16) -	(12) 33.12	(6) 63.66	(12) 263.81	(19) 6.05	(6) 26.75	(18) 46.57	(1) 0.66	(13) 14.76	(5) 5.47			(15) 4.56
	ıta Negra	SY B	peal		4.52) 1.99	1.02) 2.77	- (1.71	0.39	0.51	1.05) 0.33	0.69		0.51) 0.12	1.32	16.92		0.25) 19.57	51.11) 226.37) 7.37) 20.91	20.13) 0.70) 9.30	3.99) 11.99
	Tir	3Y B				10 (10	< 1	7 (11	.8 (16	01 (8)	61 (3)	i (6)	6 (4)	61 (12	.5 (5)	ı	(7) 15	ı	0 (33	8 (2)	56 (4)			.48 (16	.83 (9)	6.99 (14	6 (24	H (32	.56 (S)	87 (26	3 (10	8 (9)			2 (16
dry		B			(3) - (7)	(2) 1.4	(14) -	(8) 2.1	(5) 0.1	6) 0.	(10) 0.5	(6) 0.3	(10) 0.6	0.0 (6)	(10) 1.1	(5) -	(7) 0.3	(4) -	(24) 0.2	(13) 1.0	·6 (2)		(18) -	(9) 14	(12) 45	(13) 14	(11) 5.2	$(2) 3.^{\prime}$		(14) 0.3	(3) 3.7	(12) 5.9			(15) 1.7
	Sercial	101) 7.33 () 0.37) 1.56	1.12	1) 2.26	5) 1.17) 22.24	0.59	2) 0.58	12.07	t) 0.95	1.17	0.55) 19.14) 1.84	3) 0.44	5) 0.92) 74.32		3.04	t) 35.82	0) 84.93	3) 482.10	5) 25.82	38.60) 55.69	06.0	16.07	6.52			3.94
		$SY \ B^{\varepsilon}$			64 17 17	87 (4)	4 (9)	16 (1	37 (1	45 (5	60 (7	82 (1:	24 (6	32 (1	32 (5	29 (6	14 (1)	93 (6	26 (1)	34 (1	1.35 (2)		,	.88 (1)	.98 (1)	1.81 (1)	0.71 (2)	.46 (6	(I) (I)	13 (4	.89 (9	6) (6]			43 (6
,			R.I. Lit. ^d		- 0.0	1069 1.	1091 1.	1092 2.	- 0.	1186 4.	- 0.0	- 1.3	1281 3.	1319 0.	1360 1.	- 0.		- 0.9	- 0.	-	54			- 35	- 13	- 20	- 19	859 25	- 58	- 1.	- 19	- 12			- 5.
			l. R.I. c ^b Lit. ^c		2 980 6 1001	8 1053	7 1072	0 1091	5 1122	3 1178	2 1191	0 1220	8 1287	0 1323	1 1364	3 1371	9 1407	9 1410	6 1574	4 1745			. 606	670	710	705	759	865	1	904	,				- 945
			ed R.] s Cal		100	106	108	105	111	118	120	124	130	134	137	138	140	141	159	175			634	64S	969	969	712	857	902	916	920	924			924
			Compounds previously identifi in Madeira wine		59)	60)		60)		59, 60)		4, 11)	60)	11, 60)	59, 60)	60)		11, 59, 60)	11, 12, 60)										4, 10, 11)		4, 10)				
					<u> </u>	G	´ ı	J		0	'	Ċ	C	\smile		J	'	\smile	<u> </u>				'	'	'		'		43).	'	7, 43) ⁱ (.	'			'
			ical groups									1											oxyethane		,3-dioxolane	ne	ane	(9) ⁱ ethyl-propane	= 117, 103, 57, 4	ne	z = 117, 103, 57	hyl-1,3-dioxan-		(5, 43) ⁱ	ro-2H-pyran
			Chemi	Volatile Phenols	Phenol <i>o</i> -Methvlanisole	o-Cresol	<i>p</i> -Cresol	o-Guaiacol	<i>p</i> -Ethylanisole	p-Ethylphenol	p-Methylguaiacol	2-Phenoxyethano.	<i>p</i> -Ethylguaiacol	<i>p</i> -Vinylguaiacol	Eugenol	p-Propylguaiacol	Methyleugenol	Vanillin	Ethyl vanillate	Nonylphenol	Subtotal	Acetals	1-Ethoxy-1-meth	Diethoxymethane	2,4,5-Trimethyl-1	1,1-Diethoxyethaı	2-Methyl-1,3-diox	(m/z = 101, 87, 5 1,1-Diethoxy-2-m	cis-Dioxane (m/z	1,1-Diethoxybuta	cis-Dioxolane ($m_{/}$	2,6-Diethyl-5-met	4-yl acetate	(m/z = 141, 99, 5)	2-Ethoxytetrahydi
			$^{2}t_{R}^{a}(s)$		1.74 0.94	4.86	5.78	2.47	0.92	4.62	2.06	3.18	1.81	2.50	1.98	1.64	1.2.1	4.71	2.74	2.15			0.60	0.50	0.61	0.54	0.68	0.51	1.77	0.54	2.46	0.58			0.63
			$r^{-1}t_{R}^{a}(s)$		354 378	414	432	438	462	516	534	558	606	636	666	678	702	708	840	942			108	114	138	144	150	228	252	270	270	276			276
			Peak numbe		39 43	5 5	58	62	68	74	76	77	81	87	90	92	94	95	66	102			б	4	S	9	~	= :	14	15	16	18			19

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Table 1. Continued

								þ	ry				med	lium dry		
							Sercial			Tinta Negra		Ve	rdelho		Tinta Neg	ra
							SY B ^e 1	0Y B	3Y B	SY	B SY	B 10	YB 1	0Y ^f 33	(B	SY B
Peak number	$^{1}t_{R}^{a}(\mathrm{s})$	$^{2}t_{\mathrm{R}}^{a}(\mathrm{s})$	Chemical groups	Compounds previously identified in Madeira wines	R.I. R.I. Calc ^b Lit. ^c	R.I. Lit. ^d				pe	ik area g ($ imes$ 10) ⁵) (RSD ^h %				
22	288	2.89	trans-Dioxolane ((4, 10, 11)	937 -	- 12.5	3 (13) 12.61	l (5)	4.69	(8) 6.38	(15) 11.37	(12) 7.25	(8) 11.31	(17) 3.20	(11) 5.64	(19)
			$(m/z = 117, 103, 57, 43)^i$													
23	294	0.57	2-Ethoxytetrahydro-4-methyl-2H-pyran - $(m/z = 129, 85, 55, 43)^{i}$		942 -	- 1.77	(8) 4.30	(5)	0.16	(16) 0.25	(25) 1.67	(7) 1.34	(20) 1.79	(15) 6.39	(22) 5.25	(12)
24	294	1.26	2,2-Diethoxyethanol		942 948	- 1.59	(7) 1.43	(2)	0.93	(12) 1.10	(3) 2.14	(6) 0.38	(18) 1.57	(15) 0.95	(6) 1.86	(17)
28	318	0.53	1,1-Diethoxy-3-methylbutane		965 959	953 64.9	2 (16) 74.47	7 (2)	18.01	(9) 44.79	(3) 50.60	(5) 36.24	(5) 54.00	(7) 24.57	(7) 35.6	3 (8)
30	324	0.82	2-Ethyl-5-methyl-1,3-dioxane		971 919	- 0.56	(18) 0.45	(5)		0.30	. (4)	- 0.38	(7) 0.49	(12) -	1 1	ı
			$(m/z = 129, 101, 55, 41)^i$													
31	330	0.54	1-(1-Ethoxyethoxy)-pentane		977 1004	970 65.5	5 (10) 67.62	(9) +	23.00	(10) 43.67	(15) 52.02	(9) 45.75	(9) 67.52	(7) 30.05	(9) 35.2	2 (8)
32	330	0.65	4,5-Dimethyl-1,3-dioxane-5-methanol		- 976	•	•	,		•	•	•	- 5.66	(4) -	•	
			$(m/z = 113, 101, 71 55, 43)^{i}$													
33	330	4.09	trans-Dioxane $(m/z = 117,$	(4, 10, 11)	- 086	- 38.8	4 (16) 53.57	(9)	10.82	(10) 27.07	(11) 41.16	(20) 40.62	(15) 47.96	(5) 12.01	(11) 19.0	9 (4)
			$103, 57, 43)^i$													
40	354	0.55	1,1-Diethoxypentane		1001 1004	995 2.26	(17) 2.43	(19)	0.95	(8) 1.34	(9) 1.53	(18) 1.19	(7) 1.73	(19) 1.09	(23) 1.07	(17)
46	390	0.69	2-Propyl-1,3-dioxolane		1038 -	- 8.88	(3) 6.40	(18)	2.47	(2) 2.91	(5) 7.46	(11) 7.47	(1) 7.35	(15) 3.31	(10) 5.56	(19)
			$(m/z = 115, 73, 71, 45)^i$													
59	432	0.62	1,3,3-Triethoxypropane		1082 1076	- 6.30	(6) 5.11	(6)	0.94	(17) 1.51	(15) 3.18	(10) 2.71	(8) 3.62	(13) 1.11	(19) 2.24	(2)
64	444	0.56	1,1-Diethoxyhexane		1094 1092	1092 8.53	(7) 7.45	(5)	2.25	(8) 4.83	(13) 6.39	(12) 4.71	(5) 5.27	(15) 7.22	(6) 9.06	(8)
99	456	0.57	1-(1-Ethoxyethoxy)-hexane		1108 1103	- 4.20	(22) 1.19	(4)	0.76	(21) 2.05	(14) 1.53	(14) 0.98	(10) 1.69	(2) 1.36	(13) 1.16	(10)
69	462	0.62	cis-1,1-Diethoxy-3-hexene		1115 1111	- 1.60	(11) 2.24	(5)	0.44	(18) 0.34	(6) 1.57	(9) 1.24	(6) 0.93	(3) 0.90	(10) 1.24	(18)
75	528	0.58	1,1-Diethoxyheptane		1192 1202	- 1.27	(5) 1.52	(4)	0.38	(11) 1.11	(14) 0.91	(14) 0.59	(7) 1.02	(9) 0.88	(5) 1.11	(14)
82	612	0.58	1,1-Diethoxyoctane		1313 1302	- 1.43	(13) 1.61	(1)	0.99	(6) 2.18	(13) 1.29	(16) 0.89	(3) 0.61	(17) 1.23	(17) 1.58	(8)
84	618	1.07	4,5-Dimethyl-2-phenyl-1,3-dioxolane		1320 -	- 1.47	(1) 2.63	(9)	0.41	(11) 0.77	(6) 0.91	(12) 4.44	(17) 1.21	(15) 0.44	(4) 0.50	(7)
			$(m/z = 196, 152, 77, 43)^{i}$													
93	684	0.59	1,1-Diethoxynonane		1388 1401	1381 4.79	(16) 7.72	(14)	2.80	(12) 5.18	(10) 5.42	(11) 1.96	(19) 2.28	(16) 3.45	(19) 7.16	(2)
98	762	0.60	1,1-Diethoxydecane		1485 1501	- 2.17	(12) 5.25	(14)	1.85	(18) 3.99	(11) 2.20	(16) 0.93	(11) 0.97	(7) 4.69	(5) 4.56	(11)
			Subtotal			630.	91 (4) 1011	.44 (7)	319.19	(7) 521.45	(7) 656.78	(11) 707.6	5 (6) 815.0	7 (2) 529.3	3 (9) 560.	04 (2)
^a Retenti	on tim	nes in se	econds (s) for first $\begin{pmatrix} ^{1}t_{R} \end{pmatrix}$ and secon	d $\binom{2}{t_{\rm R}}$ dimensions.	'RI: reten	tion inde	x obtained th	nrough th	ne mod	ulated chro	matogram.	RI: reten	tion index r	eported in th	ne literatur	e for one
dimensic	nal G	C with	a 5%-Phenyl-methylpolysiloxane C	C column or equiva	lent. ^{34–48}	^d RI: rete	ention index	reported	l in the	literature f	or a compre	hensive G	$C \times GC$ sy	stem with E	Iquity-5 for	the first
dimensic	'n.	^{51 e} YB:	blend wine; years, average age. ^J Y: v	rintage wines; age ex	ressed in	years. ^g M	ean of three	replicate	s. ^{<i>h</i>} Rela	tive standa:	rd deviation	expressec	in percenta	ge. ⁱ Tentativ	vely identif	ied based
on mass	spectr	ra.)									,)		

Table 2.	Volati	ile Co	mpounds Identified	by HS-SPME	C/GC	×GC	-ToFl	MS in Sv	veet (Ma	lvasia ar	nd Tinta	Negra)	and Me	lium Sw	eet (Bual	and Tinta	Negra) I	Aadeira V	Vines
									sweet						me	dium sweet			
							A	Ialvasia			Tinta N	egra			Bual			Tinta Negra	-
					SY B ^e	10Y	B	SY B	18Y ^f	20Y	3Y B	SY B	SY B	10Y B	1SY B	17Y 1	9Y 3)	'B SY	В
Peak numl	$\operatorname{ver}^{1}t_{R}^{a}(s)$	$^{2}t_{R}^{a}(s)$	Chemical groups	R.I. R.I. R.I. Calc ^b Lit. ^c Lit. ^d							peak	area ^g (\times 10	$^{5})$ (RSD ^h %	~					
-	84	I 046 F	Furans $(m/z = 68, 30)^i$	55 065		62.0	(y) 156	6 (13) 294	۲ (17) ۲	S (17) -	- 23	- (14) -	-	33 (16)0	20 (2) 086	(24)043	- (6)	- 100	(15)
6 67 80	108	0.50 ¹ 1.80 3	Fetrahydrofuran 5-Furfural	634 623 - 4 803 832 - 1	73 (7 07 (1) 4.95 2) 1.22	(5) 9.28 (8) 1.98	(15) 3.28 (15) 3.28 (6) 2.40	$\begin{array}{c} (5) \\ (6) \\ (2) \\ 2.09 \end{array}$	(7) - (7) (7) - (7)		2 (3) - 6 (9) 1.(4. (1) 1.	12 (17) - 13 (3) 14	- 6.46 7 (4) 1.61	(18) - (10) 1.63	 (14) 0.45	(13) 0.87	<u>(</u>
9 10	204 216	0.61 2 1.90 2	2-Ethoxytetrahydrofuran 2-Furfural	812 825 - 1 837 829 838 3	1.43 (1 87.66 (1	1) 22.29 2) 609.07	(19) 24.2 (13) 453.	2 (2) 27.7 24 (6) 502	3 (7) 28.3 77 (17) 444	6 (9) 28. .84 (16) 482	73 (9) 16. 71 (10)53 ²	17 (9) 16 1.85 (15) 46	.37 (10) 26 3.75 (14) 54	.96 (13)21 4.90 (5) 54	.55 (19) 23.8 9.06 (9) 416	9 (16) 38.98 76 (12) 571.2	(9) 11.03 8 (6) 442.4	(8) 22.19 9 (10) 612.45	(II) (II)
12	240	3.75 2	2-Furanmethanol	874 866 862 2.	06 (1	9) 3.92	(5) 32.5	6 (9) 13.7	9 (10) 13.3	9 (14) 3.3	1 (18)9.8	2 (11) 2.3	32 (9) 3.	Н6 (11)2.8	6 (17) 10.1	6 (4) 3.62	(8) 1.96	(16) 2.75	(15)
17	276	1.65 2	2-Acetylfuran	925 911 915 30	0.01 (6	() 41.47	(5) 50.4	3 (8) 58.8	6 (4) 69.3	4 (2) 21.	25 (5) 32.	68 (3) 20	.80 (1) 24	.64 (7) 36	.35 (10) 60.9	2 (4) 27.72	(10) 16.66	(11) 24.27	(4)
29	324	1.68 5	5-Methyl-2-furfural	972 945 968 9.	5.02 (1	4) 117.19	(7) 124.	711 (11) 717	85 (6) 144	81 (2) 51.0	03 (4) 64.	52 (5) 62	.91 (4) 60	11 (4) 11	5.93 (15) 205	28 (5) 95.63	(13) 42.60	(4) 68.84	(3)
35 37	348 354	0.65 2 1.12 F	2-Pentylfuran Arhvi 2-firroate	995 993 - 4. 1001 1009 1000 1	.39 (2 1.30 (1	.) 2.98 2.) 14.90	(18) 3.49 (7) 16.7	(11) 6.33 (2) 23.7	(12) 4.34 8 (5) 21.9	- (6) 7.3. 2. (3) 4.6	2 (14)2.3 9 (9) 74	3 (18) 5.6 3 (8) 7.6	10 35 10 35 11 (2) 11)3 (6) 4.0 83 (3) 14	7 (11)3.7(48 (15)19.6	(13)4.22 8 (7) 22.76	(4) 4.64 (6) 3.16	(8) 5.65	(6)
38	354	1.22 E	Jenzofuran	1001 1005 1001 0	9) 66) 1.16	(6) 0.99	(e) 0.66	(6) 0.95	(4) 1.3	0.8 (4) 0.8	4 (10) 1.1	(1) (4) (4) (4) (4) (4) (4) (4) (4) (4) (4		(0) 0.97	(5) 0.90	(1) 1.16	(2) 1.24	(9)
42	366	1.42 1	l-(2-Furyl)-1-propanone	1014 1008 10114	.15 (1	3) 9.00	(6) 6.17	(12) 10.1	7 (7) 8.97	(3) 2.13	2 (3) 3.1	4 (5) 2.7	75 (2) 2.	37 (4) 4.9	2 (17) 10.0	1 (8) 5.63	(14) 2.19	(2) 2.47	(5)
45	390	1.45 2	2-Acetyl-5-methylfuran	1037 1039 1039 3.	86 (1	0) 6.39	(5) 5.38	(6) 6.75	(9) 8.19	(5) 1.80	3 (9) 1.9	6 (9) 3.1	[2 (3) 2.	22 (6) 8.(5 (15) 15.9	9 (12) 5.92	(17) 1.75	(13) 2.73	(9)
51	408	1.44 5	5-Ethyl-2-furfural	1058 1032 - 0.	86 (1	4) 3.67	(12) 2.87	(21) 3.73	(10) 0.99	(17) 0.4	1 (12)-	- 0.4	H5 (8) 0.	6 (11) 2.0	5 (14) 2.25	(19) 2.86	(18) 0.40	(2) 0.53	(1)
56	432	0.83 2	2-Diethoxymethylfuran	1082 1078 - 2.	82 (1	3) 6.11	(16) 7.26	66.7 (6)	(13) 7.12	(4) 2.80	0 (14)3.4	5 (13) 3.5	57 (10) 4.)5 (10) 5.0	3 (13) 9.07	(9) 9.27	(5) 1.83	(10) 3.18	(16)
57	432	3.94 5	5-Formylfurfural	1086 0.	98 (7) 17.27	(10) 13.8	5 (19) 12.5	4 (6) 9.29	(16) 4.9	5 (14)9.0	5 (12) 10	.06 (11) 7.	57 (4) 10	.20 (12) 9.62	(4) 24.74	(10) 4.65	(18) 9.03	(6)
			$(m/z = 124, 123, 95)^i$																
60	438	1.48 3	3-Acetyl-2,5-dimethylfuran	1089 1103 - 1.	36 (5) 2.40	(6) 1.79	(10) 1.36	(11) 2.37	(11) 0.4-	4 (7) 0.4	9 (10) 0.8	37 (3) 0.	1 (19) 1.	0 (9) 3.15	(8) 1.19	(8) 0.32	(8) 0.48	(14)
61	438	4.04 I	Methyl 2-furoate	1092 1081 - 2.	.15 (1	3) 1.48	(13) 3.62	(2) 1.27	(4) 1.94	. (4) 1.7	5 (19)1.7	8 (2) 1.0	10 (2) 07	6 (9) 0.8	32 (18) 2.19	(3) 1.98	(19) 2.15	(9) 1.44	(6)
63	444	1.33 I	Furaneol	1095 1090 - 0.	93 16	1.10	(13) 2.33	(1) 2.74	(8) 2.44	(4) 0.4	1 (6) 0.6	0 (2) 0.8	33 (10) 1.	06 (13) 1. ⁴	9 (8) 1.07	(7) 1.67	(6) 0.25	(11)-	
65	456	1.09 2	2-Methylbenzofuran	1108 1131 - 1.	28 18	1.67	(17) 1.16	(18) 0.67	(16) 1.06	(18) 0.7:	5 (18)0.5	1 (7) 1.	[3 (18) 0.	2 (12) 1.3	6 (19) 1.73	(17) 0.96	(13) 1.21	(14) 0.85	(3)
67	462 、	3.09 I	Maltol	1118 1110 - 0.	52 18	1.21	(20) 0.78	(17) 0.48	(20) 0.38	(18) 0.40	0 (14)0.5	1 (4) 0.3	31 (5) -	- 0.7	8 (3) 1.23	(3) 0.82	(17) 0.40	(13) 0.52	(15)
70	480	1.22 2	2-Methyl-5-propionylfuran	1136 1151 - 0.	44 9	0.64	(5) 0.31	(12) 0.60	(3) 0.68	(17) 0.1	3 (8) 0.0	9 (21) 0.2	21 (13) 0.	15 (8) 0.5	2 (4) 0.30	(5) 0.49	(9) 0.12	(17) 0.16	(14)
78	564	1.69 5	5-Ethoxymethyl-2-furfural	1247 7.	.13 9	37.56	(9) 23.7	1 (11) 22.2	9 (8) 43.3	6 (6) 8.1:	5 (7) 7.4	8 (8) 11	.40 (11) 6.	88 (6) 71	.77 (10) 22.7	5 (4) 40.27	(9) 8.60	(5) 12.87	(11)
			$(m/z = 126, 109, 81, 53)^{i}$	•													,		
79	588	2.64	5-Hydroxymethylfurfural	1286 1256 - 3.	32 26	23.50	(7) 32.5	4 (13)16.6	2 (5) 13.7	5 (15)6.4	3 (7) 14.	88 (11) 5.2	20 (3) 9.	35 (17) 12	.29 (11) 17.8	8 (19)26.33	(4) 3.73	(2) 13.98	(18)
85	624	2.92	5-Hydroxymethyldi- 116	1328 1	2.29 10	9.34	(8) 21.0	1 (11) 14.6	4 (13) 23.6	4 (10) 17.	72 (11)18.	75 (5) 16	.53 (9) 11	.99 (15)24	.04 (5) 30.1	8 (9) 26.44	(15) 10.98	(7) 20.73	(15)
86	630 (0.99 2	2-one (m/z = 85, 57, 29) ⁻ 2-Methyl-2,3-	1332 1385 - 0	64 (9) 0.56	12 0.45	(9) 0.21	(2) 0.23	(7) 0.8	5 (1) 0.4	9.1 (61))6 (8) <u>0</u> .	20 (16) 0.6	32 (6) 0.52	(14) 0.50	(19)0.64	(16) 0.76	(5)
	I		dihydrobenzofuran	1		~		~			~		~	~					Ì
			(m/z = 134, 119,																
			91, 39) ⁱ																
			Subtotal	Š	91.38(7) 941.83	(7) 856.	57(3) 888	45 (11) 869	.24(9) 650	.41(8) 76	5.80(10)64	0.36(9) 73	3.54(5) 89	3.22 (9) 878	21(8) 916.2	3 (2) 563.3	6(7) 815.75	(8)

Continued
d.
Table

							swee	ţ						mediu	m sweet			
						Malvasia				Tinta Negra			н	hual			Tinta Negr	e
				$SY B^{e}$	10Y B	15Y B	18V^{f}	20Y	3Y I	B SY	B SY	B 10Y	B 15Y	B 17	Y 19'	Y 3Y	B SY	В
Peak number ¹ t _R ^a (:	$s)^{2}t_{R}^{a}(s)$	Chemical groups	R.I. R.I. R.I. Calc ^b Lit. ^c Lit. ^d							peak area	$^{g}(\times 10^{5})$ (:	RSD^h %)						
		Lactones																
13 240	4.90	2,5-Furandione	881 828	- 1	.36 (9)	1.58 (7)	1.10 (7	,) 1.03 (- (8		- 2.59	(3) 1.45	(25) 1.00	(26) 0.83	(16) 2.01	(11)-		
)	$(m/z = 98, 54, 26)^i$																
20 282	2.53	γ -Butyrolactone	932 915 920 28.	.61 (18) 2	(2) (6)	33.22 (7)	40.74 (9) 58.95 (6) 28.04	(16) 27.34	(2) 27.81	(12) 23.72	(9) 55.78	(10) 88.52	(13) 29.00	(8) 18.53	(4) 35.00	(6)
21 282	4.15	γ -Crotonolactone	933 915 - 7.3	37 (1) 6	36 (10)	15.37 (16) 3.62 (1	2) 5.62 (13) 3.90	(18)4.53	(12) 4.37	(6) 2.63	(15) 3.71	(9) 4.99	(13)7.14	(4) 2.80	(7) 3.78	(10)
25 306	2.67	α-Angelicalactone	956 920 - 1.1	12 (20) 1	.08 (2)	1.87 (6)	8) (8	() 1.34 (10) 0.96	(7) 0.83	(7) 1.60	(22) 1.22	(7) 0.91	(4) 0.52	(5) 0.75	(10) 0.80	(13) 1.17	(12)
		$(m/z = 98, 55, 43)^i$																
26 306	2.99	3-Methylenedihydro-	956 1.0	9 (16) 3	(73 (2)	3.67 (19) 1.45 (1	2) 1.52 (22) 0.24	(17) 2.44	(9) 4.71	(18) 4.40	(4) 6.47	(15) 6.35	(8) 6.73	(5) 1.15	(22) -	,
	. 4	2,5-furandione $(m/z =$																
		$112, 84, 68, 40)^i$																
27 318	1.98	γ -Pentalactone	967 943 - 1.8	33 (13) 1	.11 (15)	(6) (9)	1.55 (9) 2.19 (3) 0.94	(13)0.51	(10) 0.64	(11) 0.73	(14) 1.77	(7) 2.74	(10) 1.11	(5) 0.32	(14) 0.73	(13)
34 342	2.55 (α -Methyl- γ -crotonolactone	991 979 - 0.5	57 (14) 0	1.56 (11)	4.64 (8)	8.52 (9) 18.43 (5) 4.82	(14) 2.50	(8) 3.91	(19) 10.27	(4) 11.06	(13) 22.63	(17) 1.14	(15) 1.00	(3) 2.84	(18)
36 348	2.99	2H-Pyran-2-one	997 967 - 0.2	12 (26) 0	125 (13)	0.41 (12	0.32 (1	(6) 0.40	10)-	- 0.66	(17) 0.22	(3) 0.20	(13) 0.19	(15) 0.33	(17) 0.37	(9) 0.17	(13) -	ı
41 360	1.52 /	$eta_{*}eta$ -Dimethylbutylrolactone	1008 992 -	- 0	1.22 (28)		0.20 (1	(4) 0.35 (20)-		- 0.19	(31) -		- 0.42	(19) -	1	1	1
44 384	4.38	3,4 Dihydro-3-methyl-2,5-	1036 1057- 0.2	36 (18) 0	1.35 (4)	4.73 (11) 2.06 (1	(9) 2.29 (16) 1.29	(4) 3.18	(13) 4.71	(7) 2.74	(6) 2.63	(8) 2.22	(19) 2.09	(9) 0.28	(13) 1.82	(4)
	Ŧ	furandione																
47 396	1.66	Lavander lactone	1046 1039 1045 0.3	31 (11) 0	0.41 (14)	0.33 (18	0.56 (1	8) 0.33 ((33) 0.35	(13)0.24	- (8)		- 0.42	(17) 0.47	(18) 1.08	(2) -		
48 396	2.02	3,4 Dimethyl-2,5-furandione	1046 0.6	57 (11) 1	.51 (3)	1.38 (12) 2.43 (1	0) 1.27 (9) 1.13	(16) 0.68	(8) 0.95	(13) 1.05	(5) 1.42	(19) 1.97	(10) 2.13	(16) 0.38	(7) 0.99	(17)
	-	$(m/z = 126, 82, 54, 39)^i$																
49 396	5.90	Pantolactone	1050 - 10600.8	37 (10) 1	.02 (4)	3.08 (23	3.09 (3	3) 2.88 (8) 2.25	(9) 2.63	(17) 1.10	(24) 1.26	(9) 2.53	(14) 4.28	(16) 2.81	- (8)	- 1.65	(12)
	-	$(m/z = 131, 71, 57, 43)^i$																
50 402	3.30	4-Methyl-2(5H)-furanone	1053 1036-	- 0	0.60 (12)	1.09 (16	0.52 (1	() 0.70 ()	24) 0.40	(18) -	- 0.50	(17) 0.40	(8) 0.53	(6) 0.52	(11) 1.27	(32)-	•	
52 408	1.76	γ -Hexalactone	1058 1055 1060 0.6	9 (9) 05	0.74 (3)	0.83 (7)	0.75 (5	;) 1.04 (1) 0.89	(16) 0.47	(4) 0.89	(20) 0.65	(13) 1.03	(11) 1.24	(9) 0.79	(13) 0.36	(9) 0.97	(11)
53 408	: 2.43 /	eta -methyl- γ -butyrolactone	1059 - 2.7	74 (5) 3	1.74 (2)	3.51 (9)	3.29 (1	9) 4.12 (11) 2.83	(4) 2.91	(8) 0.80	(11) 2.45	(15) 3.00	(19) 2.83	(11) 4.24	(16)-	- 3.08	(5)
	-	$(m/z = 85, 56, 41)^i$																
55 420	1.75	γ -Ethoxybutyrolactone	1071 1067 1069 0.5	56 (30)1	2.15 (19)	12.51 (11) 12.22 (1	1) 15.19 (8) 5.92	(14) 5.91	(11) 6.70	(14) 6.79	(18) 13.54	(11) 16.08	(2) 18.71	(14) 1.00	(2) 4.54	(19)
71 480	3.87	Solerone $(m/z =$	1140 1107- 0.2	27 (10) 0	123 (12)	0.47 (20) 0.25 (8	3) 0.43 (10) 0.30	(6) 0.34	(5) 0.43	(15) 0.41	(4) 0.53	(9) 0.61	(2) 0.50	(5) 0.23	(1) 0.57	(9)
		118, 56, 41) ⁱ																
72 492	2.43	γ -Heptalactone	1152 1130 1162 0.4	t7 (20) 0	1.92 (16)	0.85 (18) 1.08 (3	() 1.59 (12)-	- 0.26	(30) 0.24	(19) 0.27	(9) 1.22	(15) 1.94	(6) 1.03	- (2)		
73 510	2.86	Mevalonic lactone	1174 1156- 0.5	77 (18) 1	.34 (17)	1.46 (5)	0.97 (1	8) 1.85 (10) 0.59	(10)0.54	(8) 0.65	(4) 0.43	(7) 2.24	(25) 2.78	(3) 2.06	(15) 0.36	(10) 0.68	(9)
80 588	1.50	γ -Octalactone	1284 1287 1291 0.6	51 (19)0	1.95 12	0.67 (3)	0.47 (5	0.73 (12) 0.26	(3) 0.28	(6) 0.58	(4) 0.32	(14) 1.06	(16) 0.81	(15)0.65	(14) 0.40	(6) 0.60	(12)
83 618	1.26 1	<i>trans</i> -oak-lactone	1320 1292 12977.6	6 (9) 8	(7) 06.	14.55 (5)	10.77 (2) 16.94 (7) 1.42	(14) 2.90	(7) 1.16	(11) 5.52	(8) 15.81	(17) 28.34	(8) 16.09	(16) 0.29	(19) 3.27	(3)
88 642	1.37 6	cis-oak-lactone	1345 1340 1329 20.	.80 (11) 3	(5) (8)	62.53 (7)	62.57 (6	6) 69.31 (6) 4.12	(5) 8.77	(2) 24.14	(12) 32.01	(8) 86.41	(15) 95.26	(3) 80.70	(13) 0.91	(17) 9.50	(8)
89 660	3.39	2-Benzofuran-1(3H)-one		- 0	121 23	0.30 (9)		0.31 (20) 0.20	(11) 0.16	(22) 0.20	(16) 0.16	(11) 0.48	(13) 0.31	(15) 0.37	(11) 0.22	(13) 0.21	(8)
	-	$(m/z = 134, 105, 77, 51)^{i}$																

Continued	
Table 2.	

			1			SW	reet						mediu	n sweet			
					Ma	lvasia			Tinta Negra			B	lai			l'inta Negra	
				5Y B ^e 10	Y B 15	Y B 18	γ ^f 201	Y 3Y	B SY	B SY	B 10Y	B 15Y	B 177	(91 J	r 3Y]	SY F	
Peak numbe	$r^{1}t_{R}^{a}(s)^{2}t_{R}^{a}(s)$	chemical groups	R.I. R.I. R.I. Calc ^b Litt. ^c Litt. ^d						peak area	$^{g}(\times 10^{5})($	RSD^h %)						
91	672 1.44	γ -Nonalactone	1377 1366 1368 2.59) (20) 3.22	(13) 1.74	(6) 1.16	(6) 1.85	(19) 2.26	(9) 0.77	(18) 3.07	(10) 1.25	(9) 4.39	(9) 1.92	(19) 2.13	(15) 1.59	(17) 1.75	(16)
96	750 1.40	γ -Decalactone	1471 1472 1472 0.63	3 (13) 1.07	(9) 0.36	(10) 0.33	(9) 0.60	(11) 0.78	(10) 0.15	(20) 0.14	(14) 0.31	(3) 0.68	(5) 0.56	(25) 0.49	(16) 0.47	(13) 0.45	(6)
97	762 1.64	Massoia lactone	1487 1508 1482 0.27	, (16) 0.30	(13) 0.26	(10) 0.13	(7) 0.25	(10) 0.18	(14) 0.07	(14) 0.20	(13) 0.09	(10) 0.55	(7) 0.38	(13) 0.23	(24) 0.11	(10) 0.17	(2)
100	900 1.25	γ -Dodecalactone	1684 1685 - 0.13	(9) 0.11	(21) 0.19	(9) 0.17	(11) 0.23	(14) 0.57	(20) 0.10	(18) 0.33	(25) 0.11	(13) 0.26	(19) 0.10	(23) 0.32	(17) 0.99	(20) 1.23	(2)
101	918 1.33	δ -Dodecalactone	1713 1721 - 0.43	(21) 0.35	(16)-			- 0.64	(32)-	- 0.23	(23) -			- 0.23	(21) 0.65	(59 0.62	(2)
103	996 0.87	Muskolactone	1841 1839- 1.73	(19) 3.14	(2) 1.35	(14) 0.64	(18) 0.94	(8) 0.97	(7) 1.03	(4) 1.39	(12) 0.54	(2) 0.67	- (9)	- 0.95	(16)-	2.43	(17)
		Subtotal	83.	73 (10) 122.4	46 (2) 173.5	6(5) 161.93	3(4) 212.71	(5) 66.24	(8) 70.19	(3) 94.45	(10) 101.37	(6) 220.29	(8) 289.95	(5) 187.16	(11) 33.02	(3) 78.06	(5)
		Volatile Phenols															
39	354 1.74	Phenol	1002 980 996 1.03	(15) 0.93	(12) 1.68	(11) 1.98	(15) 1.85	(8) 5.47	(11) 3.60	(4) 5.33	(6) 5.05	(16) 8.35	(6) 3.69	(12) 4.77	(11)-	6.34	(8)
43	378 0.94	o-Methylanisole	1026 1001 - 0.35	(4) 0.43	(5) 0.21	(22) 0.17	(13)-	- 0.11	(19)-	- 0.30	(9) 0.48	(6) 0.32	(10) 0.25	(7) 0.39	(10) 0.10	(19) 0.23	(13)
54	414 4.86	o-Cresol	1068 1053 1069 1.35	(6) 1.25	(18) 1.35	(3) 0.95	(10) 1.25	(6) 1.79	(7) 1.23	(12) 2.22	(11) 0.99	(14) 2.11	(17) 1.33	(14) 1.08	(2) 1.57	(7) 3.24	(2)
58	432 5.78	<i>p</i> -Cresol	1087 1072 1091 -	- 0.57	(17) 0.76	(22) 0.46	(20) 1.12	(10) 0.85	(16) 0.66	(3) 0.59	(9) 0.59	(16) 2.25	(8) 1.72	(13) 1.88	(14) -	1.31	(21)
62	438 2.47	o-Guaiacol	1090 1091 1092 2.49	(10) 2.40	(7) 3.18	(11) 1.93	(11) 2.48	(9) 2.21	(4) 1.61	(3) 2.22	(6) 1.29	(13) 3.65	(9) 4.92	(8) 1.83	(15)2.15	(5) 2.68	(14)
68	462 0.92	p-Ethylanisole	1115 1122- 0.61	(14) 1.62	(8) 1.74	(4) 0.51	(6) 1.41	(15)0.30	(12) 0.77	(15) 0.62	(8) 0.70	(7) 0.73	(13) 2.46	(18) 0.81	(18) 0.16	(10) 0.30	(4)
74	516 4.62	$p ext{-}Ethylphenol$	1183 1178 1186 3.30	(6) 2.87	(12) 5.82	(13) 5.69	(18) 7.84	(7) 0.76	(13)0.93	(11) 14.11	(11) 3.56	(4) 26.19	(11) 8.29	81) 9.45	(13)-	1.74	(13)
76	534 2.06	<i>p</i> -Methylguaiacol	1202 1191 - 0.82	(16) 0.71	(6) 0.64	(25) 0.30	(18) 0.50	(14) 0.47	(4) 0.21	(9) 0.59	(13) 0.28	(20) 0.77	(15) 1.09	(13) 0.20	(6) 0.21	(1) 0.34	(13)
77	558 3.18	2-Phenoxyethanol	1240 1220 - 1.20	(17) 2.26	(18) 0.86	(20) 0.70	(7) 1.10	(12)0.73	(13)0.58	(11) 0.50	(14) 0.40	(7) 0.51	(8) 0.50	(9) 0.43	(10) 0.52	(8) 1.61	(8)
81	606 1.81	p-Ethylguaiacol	1308 1287 1281 3.38	(10) 1.95	(12) 2.34	(16) 1.28	(2) 2.18	(12) 0.59	(16)0.84	(8) 12.21	(13) 1.12	(10) 4.16	(9) 2.37	(19) 1.62	(18) 0.42	(3) 1.11	(12)
87	636 2.50	<i>p</i> -Vinylguaiacol	1340 1323 1319 0.46	(17) 0.43	(14) 0.19	- (2)		- 0.16	(13)-	- 0.13	(27) 0.19	(8) 0.19	(15) 0.12	(3) 0.29	(14)-	0.29	(9)
06	666 1.98	Eugenol	1371 1364 1360 1.15	(18) 0.96	(12) 0.48	(19)0.19	(15) 0.76	(13) 1.13	(7) 0.32	(13) 1.73	(13) 0.41	(2) 1.10	(10) 1.45	(5) 0.36	(9) 0.77	(11) 1.13	(14)
92	678 1.64	p-Propylguaiacol	1383 1371 - 0.24	+ (14) 0.20	(2) 0.20	(17) 0.11	(26) 0.17	(10)-	•	- 0.33	(4) 0.28	(18) 0.21	(15) 0.16	(9) 0.23	(5) -		
94	702 1.21	Methyleugenol	1409 1407 - 2.47	(13) 1.13	(21) 1.55	(7) 0.55	(14) 0.87	(11) 10.95	(8) 0.72	(15) 1.42	(20) 0.45	(12) 1.22	(18) 0.87	(6) 1.92	(13) 0.78	(5) 1.11	(18)
95	708 4.71	Vanillin	1419 1410- 1.00	(16) 0.75	(11) 1.45	(3) 0.23	(19) 1.05	(21) 0.41	(17) 0.29	(8) 0.69	(6) 0.65	(8) 1.79	(14) 0.52	(7) 2.59	(19)-	0.48	(8)
66	840 2.74	Ethyl vanillate	1596 1574- 0.19	(13) 0.40	(12) 0.24	(10) 0.21	(13) 0.19	(4) 0.33	(26) 0.45	(6) 0.48	(28) 0.13	(5) 0.46	(14) 0.26	(2) 0.64	(14) 0.42	(15) 0.40	(17)
102	942 2.15	Nonylphenol	1754 1745 - 1.88	(17) 2.07	(6) 1.92	(24) 1.61	(8) 1.28	(9) 0.74	(17) 0.56	(5) 1.49	(15) 0.53	(17) 1.41	(11) 0.70	(15) 1.05	(14) 1.35	(14) 1.02	(21)
		Subtotal	21.5	92 (6) 20.93	3 (5) 24.59	(9) 16.86	(8) 24.05	(7) 27.00	(2) 12.76	(2) 44.96	(2) 17.12	(6) 55.41	(7) 30.70	(7) 29.57	(5) 8.45	(4) 23.33 ((9)
		Acetals															
3	108 0.60	1-Ethoxy-1-	634 606	•										- 4.19	(20)-		
		methoxyethane															
4	114 0.50	Diethoxymethane	645 670 - 472	0 (18) 50.18	(4) 60.91	(11) 133.25	(13) 101.16	(11) 19.02	(27) 27.09	(10) 30.31	(8) 65.74	(13) 101.72	(3) 72.52	(15) 128.05	(9) 15.33	(9) 31.63	(9)
S	138 0.61	2,4,5-Trimethyl-1,3-	690 710 - 203	6 (8) 30.65	(7) 17.34	(8) 20.39	(23) 35.08	(15) 24.96	(17) 35.08	(7) 48.84	(12) 41.74	(12) 45.71	(12) 57.03	(17) 83.86	(10) 24.32	(19) 51.69	(10)
		dioxolane															
6	144 0.54	1,1-Diethoxyethane	690 705 - 197	.47 (16) 280.6	62 (10) 236.1	9 (13) 247.55	(17) 242.89	(19) 209.75	(6) 198.09	(9) 307.04	(7) 342.64	(11) 296.07	(17) 254.76	(12) 376.16	(3) 206.89	(20) 268.47	(15)
7	150 0.68	2-Methyl-1,3-dioxane	712 759 - 5.43	(20) 7.85	(7) 18.95	(15) 39.54	(2) 54.24	(15) 1.11	(13) 4.47	(18) 1.48	(19) 12.46	(11) 81.49	(10) 31.54	(7) 32.82	(8) 1.12	(12) 5.81	(17)
		$(m/z = 101, 87, 59)^i$															

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	egra	SY B		53 (17)	28 (13)	5 (10)	5 (14)				(8)	(/1)	(8)			5 (5)	86 (9)			26 (14)			97 (16)		8 (14)	35 (7)		8 (11)	5 (14)	1 (6)	5 (11)	1 (12)
	Tinta Né	Y B		(10) 14.6	(18) 31.2	(10) 0.86	(13) 7.46				(12) 5.4.	17.6 (01)	(11) 8.07			(23) 3.05	2 (3) 48.8	(11)-		. (9) 40.	•		(3) 19.5		(18) 1.9	(9) 38.		(14) 4.25	(13) 8.45	(15) 1.5]	- 1.35	(3) 0.8i
		Y 3		(14) 13.00	(17) 6.30	(10) 0.33	(16) 0.93	- (8)		101 (0)	16.4 (2)	co.c (/1)	(19) 6.10			(12) 0.14	(5) 32.32	(1) 1.47		(3) 14.10	•		(5) 3.84		(14) 1.11	(11) 14.10		(17) 1.26	(11) 4.65	(9) 0.44	(10) -	(7) 0.34
n sweet		19		(8) 65.30	(10) 37.33	(17) 1.55	(5) 12.49	(17) 10.99			12.8 (02)	10.01 (41)	(2) 3.60			(13) 4.55	(6) 76.73	(4) 0.95		(8) 59.00			(9) 63.03		(11) 3.63	(9) 62.93		(14) 11.64	(10) 11.94	(16) 2.20	(3) 2.80	(10) 1.14
mediun	al	3 17Y		[13) 83.02	18)71.09	16) 1.68	15) 19.91	(10) 14.61			40.0 (4)	04.01 (71	13) 6.46			[12) 4.39	3) 77.20	(10) 1.56		8) 71.21			16) 72.41		(4) 3.53	[17] 75.29		[12] 2.49	13) 14.94	9) 1.56	[14] 3.74	20)2.27
	Bu	3 15Y I		10) 56.00 (18) 53.56 (4) 1.36 (19) 17.39 (21) 7.63 () /0.6 (01	0.0.6/61	9) 4.27 (25) 3.90 (11) 79.37 (16) 0.70 ()		2) 57.76 (1		14) 36.05 (10) 2.72 ()	12) 57.37 (16)9.39 (2) 6.71 (12) 2.25 (6) 1.58 (4) 1.18 (
		10Y I	${ m SD}^h$ %)	3) 37.29 (7) 16.72 (20) 1.02	6) 5.16 (6) 6.84 () 4.34 (6) 2 5 4 (6	+0.0 (+	11) 3.48 (17) 1.71 (9) 47.07 (0.28 (9) 36.96 (1		16) 15.26 (11) 1.25 (4) 33.51 (8) 4.07 (7) 5.07 (7) 1.58 (17) 0.97 (1) 0.80 (
		SY B	$(\times 10^{5}) (R)$	6) 27.93 (7) 22.34 (12) 0.48	12) 4.55 (11) 6.66		00 0 1) 60.6 (1) 61.6 (0	22) 1.35 (9) 2.26 (18) 39.29 ((2		9) 31.80 (2)		17) 16.27 (16) 0.93 (7) 15.41 (4) 3.18 (2) 4.44 (15) 1.66 (8) 1.24 (5) 0.66 (
	inta Negra	SY B	peak area ^g (13) 18.52 (4	;) 01.79 (;	17) 0.73 (3	14) 3.92 (.	14) 4.77 () 85.6 (1	00.7 (0	15) 2.16 (;			25) 2.23 8.	11) 43.00 (.	7) 0.45 (9) 29.69 (3.45 (12) 12.15 (.		10) 1.64 (.	14) 4.75 (10) 2.81 (.	9) 6.14 (;	15) 1.08 (.	8) 0.52 (3	16) 0.54 (;
	L	3Y B		4) 18.61 (3) 16.54 (16) 0.63 (14) 4.05 (10) 6.82 (L.) / (.	114	5) 1.65 ((1) 2.83 (10) 38.92 ((3) 1.01		t) 16.97 (1		(0) 11.08		10) 1.32 () 7.09 (14) 3.07 ((1) 11.31 (1)	16) 1.62 (3) 0.78 (3	10) 0.94 (
		20Y		() 82.72 (-	1) 65.87 (3	8) 1.64 (]) 16.36 ()	.) 14.46 (1		0000	2) 27.C (Q	c1.71 (n) 3.14 (5			() 7.56 ()) 61.31 ()	() 1.05 ()) 75.36 (4	1)) 51.44 ()) 2.85 ()) 77.40 (3) 5.81 ()) 15.74 ()	6) 2.39 (]) 1.53 (8) 3.13 ()
swee	ia	$18Y^{f}$		8) 86.98 (8	5) 36.43 (1	6) 1.80 (1	4) 10.14 (5) 21.85 (4		.)	1) CC.0(1 1) 01 F (0	1) (4) (1) 4.06 (3) 6.43 (8) 76.00 (6) 0.88 (8		1) 58.63 (3	6.67 (1) 39.36 (5		7) 2.51 (7	2) 58.32 (3		8) 00.6 (2) 10.52 (9) 2.74 (1) 1.48 (7) 1.03 (8
	Malvas	15Y B		1) 57.18 (3) 24.64 (1	7) 1.34 (1) 6.83 (1	8) 12.11 (6			1) 647/1 (T) 0C.C (9) 2.75 (4			5) 3.88 (5) 60.04 (7	0.87 (7) 42.45 (1	3)) 34.65 (1) 1.92 (2	0) 10.50 (1) 2.67 (5) 9.80 (1) 1.49 (7	(9) 0.10	3) 1.55 (9
		10Y B		3) 53.76 (1	() 25.15 (5	(1) 1.37	() 2.81 (5) 17.12 (1		00	0.00 (0)		5.35 (1)			2.36 (1.	55.77 (4	6) 69.0		26.44 (3	7.35 (1)) 14.09 (8		1.96 (7	14.50 (2)		(8) 1.75 (8)	6.06 (8	7.96 (5	0.63 (4	1.02 (1
		SY B^{ℓ}	_	48.55 (18	18.22 (28	0.76 (13	0.43 (14	8.11 (15			(/) 19.6	11/ 70'0	1.32 (8)			1.65 (6)	36.47 (5)	•		18.28 (7)	•		0.79 (14		1.88 (5)	12.94 (8)		1.81 (18	94.72 (8)	0.90 (5)	0.73 (6)	0.51 (7)
			R.I. R.I. R.I. alc ^b Lit. ^c Lit. ^t	57 865 859	02	19 904 -	20	24			- 046 - 16	/0	42			42 948 -	65 959 953	- 616 12		77 1004 970	2		80		001 1004 995	038		082 1076 -	094 1092 1093	108 1103 -	115 1111 -	192 1202 -
			Ps C.	1- 85	90	(7, 43) ⁱ 91	92	7, 43) ⁱ 1,3- 92		, 43) ^t T	n-pyran 9.	7 12V	, 10± (10± (10± (10± (10± (10± (10± (10±		, 43) ⁱ	94	lbutane 96	lioxane 97	S, 41) ⁱ	entane 9',	ane-5- 9',	110, 101,	36	7, 43) ⁱ	10	e 10	5) ⁱ	le I(10	exane 11	ene 11	11
			Chemical grou	Diethoxy-2-methy	ropane Vioxane	m/z = 117, 103, 5 Diethoxybutane	Jioxolane	<i>m</i> / <i>z</i> = 117, 103, 5 Diethyl-5-methyl-	lioxan-4-yl acetate	m/z = 141, 99, 55	. Dil	s-DIOXOIALIE	hoxytetrahydro-4	aethyl-2H-pyran	m/z = 129, 85, 55	Diethoxyethanol	Diethoxy-3-methy	:hyl-5-methyl-1,3	m/z = 129, 101, 5	-Ethoxyethoxy)-p	Dimethyl-1,3-diox	1 55. 43) ^{<i>i</i>}	s-Dioxane	m/z = 117, 103, 5	Diethoxypentane	opyl-1,3-dioxolan	<i>z</i> = 115, 73, 71, 4	3-Triethoxypropaı	Diethoxyhexane	-Ethoxyethoxy)-h	,1-Diethoxy-3-hex	Diethoxvheptane
			$R^{a}(s)$	51 1,1-I	Pi 77 cis-D	() 54 1,1-I	46 cis-D	() .58 2,6-I	d.	1) (1 (1)	00 Z-EU	.) (ب 57 2-Etl	Ш	<i>v</i>)	.26 2,2-I	.53 1,1-I	.82 2-Etl	r) (.54 1-(1.	.65 4,5-I	1 2	09 trans	<i>r</i>)	.55 1,1-I	.69 2-Pr	(m/;	.62 1,3,3	.56 1,1-I	.57 1-(1-	.62 cis-1,	58 1.1-I
			nber $^{1}t_{\mathrm{R}}{}^{a}(\mathrm{s})^{2}t$	228 0.	252 1.	270 0.	270 2.	276 0.			C 0/7	7 007	294 0.			294 1.	318 0.	324 0.		330 0.	330 0.		330 4.		354 0.	390 0.		432 0.	444 0.	456 0.	462 0.	52.8 0
			Peak nur	11	14	15	16	18		-	۲۱ در	77	23			24	28	30		31	32		33		40	46		59	64	66	69	75

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	Tinta Negra	3YB SYB			9 (20) 0.56 (5)			0 (14) 4.14 (16)	1 (8) 2.31 (3)	0.35(10)605.91(7)	e literature for one juity-5 for the first sly identified based
set		19Y			1.53 (13) 0.3			3.52 (11) 2.1	1.58 (16) 1.2	1082.65(4) 360	eported in the stem with Ec ge. ^{<i>i</i>} Tentative
medium swe		17Y) 1.60 (4) 1) 8.86 (16)3	0) 5.32 (16) 1	982.46(1) 1	tion index ref. $C \times GC$ sy lin percenta
	Bual	B 15Y B			(11) 2.00 (9			(9) 4.42 (7	(13) 2.36 (1	(6) 949.17(8	n. ^c RI: reten prehensive C m, expressed
		Y B 10Y		$(RSD^h \%)$	(9) 0.74			(4) 2.29	(18) 2.63	86(2) 696.06	romatogram e for a comp lard deviatic
egra	SY B S		: area g ($ imes$ 10 ⁵)	32 (13) 0.97			76 (4) 3.76	81 (11) 3.60	32.30(2) 583.	nodulated ch he literature elative stanc	
	Tinta N	3Y B		peak	- 0.			7.78 (1) 5.7	5.94 (11) 1.3	423.83(7) 43	rough the m eported in t eplicates. ^h R
t		20Y			2) 1.06 (6)			5) 14.28 (14)	2) 1.89 (6)	5) 963.25(5)	obtained thin tion index 1 an of three r
swee	lvasia	ζB 18Υ ^f			(14) 0.61 (3			(3) 4.15 (6	(9) 2.27 (5)	0(3) 869.95(ntion index ¹⁸ ^d RJ: reter 1 years. ^g Me
	Ma	10Y B 153			5 (13) 0.49			5 (17) 4.49	3 (18) 3.06	.08(6) 641.9	ns. ^b RI: rete uivalent. ^{34–4} expressed ii
		SY B ^e			.44 (16) 0.6			.74 (4) 5.2	.28 (14) 2.6	39.28(4) 637	() dimension lumn or equ e wines; age
			R.I. R.I. R.I.	Calc ^b Lit. ^c Lit. ^d	1320 - 1	152,		1388 1401 1381 2	1485 1501 - 1	4) and second $({}^{2}t_{\mathrm{f}}$ lysiloxane GC co age f Y: vintag
				Chemical groups	5-Dimethyl-2-phenyl-	-3-dioxolane (m/z = 196,)	$77, 43)^{i}$	l-Diethoxynonane	(-Diethoxydecane	btotal	nds (s) for first $({}^{1}t_{R})$ %-phenyl-methylpo nd wine; years, avera
				$\operatorname{ar}^{1}t_{R}^{a}(\mathrm{s})^{2}t_{R}^{a}(\mathrm{s})$	618 1.07 4,5	1,3		684 0.59 1,1	762 0.60 1,1	Su	t times in secon d GC with a 5 ^{49–51 e} YB: blei
				Peak numb	84			93	98		^a Retention dimension dimension.

(Figure 2). According to 2 D (polarity), it was observed that furan, tetrahydrofuran, alkyl furans (2-ethoxytetrahydrofuran and 2-pentylfuran), and furanic acetal (2-diethoxymethylfuran) exhibited similar polarity. The polarity of these furans increase by the presence of an aromatic ring $(-C_6H_5)$ (e.g., benzofuran) or a formyl (-CHO) (e.g., furanic aldehydes) group. Concerning furanic aldehydes, the presence of a second -CHO group in the α' -position related to the heteroatom of the heterocyclic ring, e. g., 5-formylfurfural (${}^{1}t_{\rm R}$ = 432 s, ${}^{2}t_{\rm R}$ =3.94 s) results in a considerable increase in polarity compared to the remaining furanic aldehydes, which were located in the chromatographic space of ${}^{1}t_{\rm R} = 198-564$ s and ${}^{2}t_{\rm R} = 1.10-1.90$ s (Figure 2). However, furanic alcohols showed a high polarity compared to furanic aldehydes; consequently, an increase was observed in ²D $({}^{2}t_{R} = 2.64 - 3.75)$. Although, Furaneol $({}^{1}t_{R} = 444 \text{ s}, {}^{2}t_{R} = 1.33 \text{ s})$ is an exception, resultant from the presence of two methyl $(-CH_3)$ groups in α - and α' -positions related to the heteroatom, a decreasing in polarity was observed. According to ²D, the elution order of furans under study based on their functional groups was alkyl \sim acetal < aromatic ring (benzofuran) < aldehyde < alcohol.

The lactones were organized in lactone/alkyl lactone, anhydride, enolic lactone, and aromatic lactone (Figure 3). The lactone/alkyl lactones are the less polar type; thus, lower ²D values (${}^{2}t_{\rm R} = 0.87 - 3.30$ s) were observed. γ -Crotonolactone $({}^{1}t_{R} = 282 \text{ s}, {}^{2}t_{R} = 4.15 \text{ s})$ is an exception; the high ${}^{2}t_{R}$ was explained due to the presence of π -bound in the structure and the absence of the $-CH_3$ group.⁵⁶ The anhydride showed higher polarity (${}^{2}t_{\rm R} = 4.38 - 4.90$ s) when compared to lactone/alkyl lactones due to the presence of RC=OR' group in the α' position instead of H. Nevertheless, two exceptions were observed, namely, 3-methylenedihydro-2,5-furandione (${}^{1}t_{R}$ = 306 s, ${}^{2}t_{\rm R}$ = 2.99 s) and 3,4-dimethyl-2,5-furandione (${}^{1}t_{\rm R}$ = 396 s, ${}^{2}t_{\rm R}$ = 2.02 s), which demonstrated lower polarity compared to others, due to the a weak π -bound in the β -position, and the two – CH₃ groups in β - and β' -positions of heterocyclic ring, respectively. Pantolactone (${}^{1}t_{\rm R} = 396$ s, ${}^{2}t_{\rm R} = 5.90$ s), the unique enolic lactone detected, showed the highest polarity of lactones, which may be explained by the presence of the -OHgroup in the β -position of the heterocyclic ring. Thus, according to ²D, the elution order of lactones under study based on their functional groups was alkyl < aromatic ring < ketone (anhydride) < alcohol.

The volatile phenols were organized in phenolic ether, alkyloxy phenol, phenol/alkyl phenol, phenolic aldehyde, phenolic ester, and phenoxy alcohol (Figure 4). According to ²D, the phenolic ether exhibited the lowest polarity (${}^{2}t_{R} = 0.92 - 1.21 \text{ s}$) compared to alkyloxy phenol, phenol/alkyl phenol, phenolic aldehyde, phenolic ester, and phenoxy alcohol. The presence of the -OH group in these former compounds increases their polarity; consequently, higher ${}^{2}t_{R}$ values were observed. Intermediate polarity was observed for alkyloxy phenol (${}^{2}t_{\rm R} = 1.64 -$ 2.50 s), followed by phenolic ester (e.g., ethyl vanillate, ${}^{2}t_{\rm R}$ = 2.74 s), phenoxy alcohol (e.g., 2-phenoxyethanol, ${}^{2}t_{R} = 3.18$ s), and phenolic aldehyde (e.g., vanillin, ${}^{2}t_{\rm R}$ = 4.71 s). The alkyl phenols, namely, o-cresol, p-cresol, and p-ethylphenol showed the highest polarity (${}^{2}t_{\rm R} = 4.62 - 5.78$ s), which may be explained by the absence of an $-OCH_3$ group in the α -position compared to all other phenols studied. Thus, according to ²D, the elution order of volatile phenols under study based on their functional groups was ether (phenolic ether < alkyloxy) < ester < aldehyde < alcohol (alkyl phenol).



Figure 2. Peak apex plots of furans identified in Madeira wine (attribution of the peak number is shown in Tables 1 and 2).



Figure 3. Peak apex plots of lactones identified in Madeira wine (attribution of the peak number is shown in Tables 1 and 2).

Concerning the acetals, three types of structures were observed: alkyl/heterocyclic acetal, alkyl/heterocyclic acetal alcohol, and aromatic acetal. Previous studies only allowed the detection of heterocyclic acetal alcohols.^{4,10,11,20} The alkyl/ heterocyclic acetals are the less polar compounds; afterward, they are the first compounds eluted according ²D. For these compounds, the maximum ²t_R achieved was 0.82 s. 2-Diethoxyethanol (alkyl acetal alcohol, ¹t_R = 294 s, ²t_R = 1.26 s) and 4,5-dimethyl-2-phenyl-1,3-dioxolane (aromatic acetal, ¹t_R = 618 s, ²t_R = 1.07 s) showed an intermediate polarity due to the presence



Figure 4. Peak apex plots of volatile phenols identified in Madeira wine (attribution of the peak number is shown in Tables 1 and 2).



Figure 5. Peak apex plots of acetals identified in Madeira wine (attribution of the peak number shown is in Tables 1 and 2).

of the -OH and $-C_6H_5$ groups. Otherwise, the four heterocyclic acetal alcohols located in the chromatographic space of ${}^{1}t_{\rm R} = 252-330$ s and ${}^{2}t_{\rm R} = 1.77-4.09$ s showed the highest polarity compared to heterocyclic acetals due to the -OH group in their structure. Thus, according to ${}^{2}D$, the elution order of acetals under study was alkyl < aromatic ring < alcohol. Regarding the peak apex of the acetals displayed in Figure 5, it is possible

to observe the advantage of the second dimension for the analysis of different acetals.

The structured 2D chromatographic profile arising from ¹D volatility and ²D polarity was observed within each chemical group based on the properties and positions of their functional groups, which allow more reliable identifications. Globally, based on the functional group of the chemical families under study, the

 ${}^{2}t_{\rm R}$ values increase in this way: alkyl < ether < ester < aromatic ring < ketones ~ aldehydes < alcohol. This information is especially useful for classifying unidentified compounds.

Establishment of Potential Age Markers. The GC peak area and RSD (relative standard deviation) values of furans, lactones, volatile phenols, and acetals obtained using HS-SPME/GC \times GC—ToFMS methodology are listed in Table 1 for dry and medium dry, and in Table 2 for sweet and medium sweet Madeira wines.

Furans. The furanic aldehydes, such as 2-furfural and 5methyl-2-furfural, were the predominant furans tentatively identified in the Madeira wines under study. Other furans were also tentatively identified, namely, 2-ethoxytetrahydrofuran, 2-acetylfuran, ethyl 2-furoate, 5-formylfurfural, 5-ethoxymethyl-2-furfural, 5-hydroxymethylfurfural, and 5-hydroxymethyldihydrofuran-2-one. Some of these furans have been reported as age markers of Madeira wine, namely, ethyl 2-furoate, 5-ethoxymethyl-2-furfural, and 5-hydroxymethylfurfural, and a similar trend during aging was observed.^{4,12} From the total of 26 furans tentatively identified, only 8 compounds were previously identified in Madeira wines (see Table 1), and as far we know, the 18 furans listed in Tables 1 and 2 are detected for the first time in Madeira wine. In order to evaluate the trend of furans, lactones, volatile phenols, and acetals with the aging process, correlations between GC peak area and age were computed. This approach was only applied to Malvasia and Bual Madeira wines, as they are the unique varieties under study that presented samples with 5 different ages. The other varieties under study only have wines with 2 different ages. A correlation (r = 0.87) between total GC peak area of furans and age was found for Bual, while a lower correlation (r = 0.56) was achieved for Malvasia, which means that, for Bual, the total GC peak area of furans was 76% related with age (expressed as r^2 (coefficient of determination)), whereas for Malvasia, it was only 31%. Furthermore, from 26 furans tentatively identified, only 2 are the highest and positively correlated with wine age, namely, 3-furfural (r = 0.88and 0.78, respectively for Malvasia and Bual), and ethyl 2-furoate (r =0.96 and 0.99). Thus, these compounds may be suggested as potential age markers for Malvasia and Bual varieties.

Lactones. γ -Butyrolactone and *cis*-oak-lactone were the main lactones detected. Regarding the two isomers of whisky lactones (cis- and trans-oak-lactones), the cis isomer showed the highest GC peak area than trans-isomer. As previously reported, in the wine acidic medium, an easier extraction of cis-oak-lactones occurs.¹⁶ γ -Lactones are among the most important compounds from a sensory point of view, and their content tends to increase during the aging process in oak casks (Tables 1 and 2). Similar trends of γ -lactones during aging were achieved for Madeira and red wines by Câmara et al.⁴ and Cérdan et al.,⁵ respectively. From the total of 30 lactones tentatively identified, only 9 compounds were previously identified in Madeira wines (see Table 1), and as far we know, the 21 lactones listed in Tables 1 and 2 are detected for the first time in Madeira wine. The total GC peak area of lactones showed a high correlation (r = 0.93) with age in Malvasia, whereas for Bual, a slightly lower correlation (r =0.71) was observed. Thus, for Malvasia the total GC peak area of lactones was 87% related with age, whereas for Bual, it was only 50%. Moreover, from the 30 lactones tentatively identified, only 5 are the highest and positively correlated with age for Malvasia and Bual varieties, namely, pantolactone (r = 0.78 and 0.77, respectively, for Malvasia and Bual), γ -ethoxybutyrolactone (r = 0.76 and 0.92), γ -heptalactone (r = 0.87 and 0.72), *trans*-oaklactone (r = 0.83 and 0.77), and *cis*-oak-lactone (r = 0.93 and 0.82).

Volatile Phenols. The predominant volatile phenols detected were o-guaiacol, p-ethylphenol, and p-ethylguaiacol. During aging, for wines obtained from the Sercial grape variety (alcoholic degree of 16.8%⁴) these compounds increase more remarkably, whereas for Bual wines $(17.8\%^4)$, a considerable decrease was observed. According to Dias et al.,⁵⁷ the content of *p*-ethylphenol and *p*-ethylguaiacol is more accentuated in wines with lower alcoholic degree because the high content reduces the microbial activity of yeast, making the synthesis of ethylphenols difficult. From the total of 17 volatile phenols tentatively identified, 11 compounds were previously identified in Madeira wines (see Table 1), and as far we know, the 6 volatile phenols listed in Tables 1 and 2 are detected for the first time in Madeira wine. Conversely, volatile phenols did not show a linear trend with age for Malvasia (r = 0.05) and Bual (r = 0.05) varieties. In addition, the individual analysis of all volatile phenols revealed that the GC peak area of these components were not correlated with wine age, with the exception of *p*-cresol (r = 0.79 and 0.75, respectively for Malvasia and Bual). This behavior may be explained as the volatile phenols present several origins, i.e., oak, microbiological activity (e.g., Brettanomyces and Dekkara), and hydroxycinnamic acids of wine,²³ among others.

Acetals. The major acetals detected were diethoxymethane, 1,1diethoxyethane, 1,1-diethoxy-2-methyl-propane, 1,1-diethoxy-3methylbutane, and 1-(1-ethoxyethoxy)-pentane. From the total of 30 acetals tentatively identified, only 4 compounds were previously identified in Madeira wines (see Table 1), and as far we know, the 26 acetals listed in Tables 1 and 2 are detected for the first time in Madeira wine. The white varieties (Malvasia, Bual, Sercial, and Verdelho) presented high total GC peak area of acetals than the red one (Tinta Negra). Similar results were achieved by Cutzach et al.,⁵⁸ as the high content of polyphenols in red varieties slowed the oxidation reaction and combined easily with acetaldehyde. For acetals, high correlation with age was observed for Malvasia (r = 0.95), and Bual (r = 0.95). Similar correlation was observed by Câmara et al.^{4,10} for heterocyclic acetals alcohols previously identified in Madeira wines. Additionally, other acetals such as 1,1-diethoxy-2-methyl-propane (r = 0.81 and 0.86, respectively, for Malvasia and Bual), 2,2-diethoxyethanol (r = 0.96 and 0.84), 1-(1-ethoxyethoxy)-pentane (r = 0.96 and 0.81), 1,1diethoxypentane (r = 0.81 and 0.94), 2-propyl-1,3-dioxolane (r = 0.86 and 0.87), and 1,1-diethoxyhexane (r = 0.95 and 0.83)showed high correlation with age for Malvasia and Bual varieties.

In sum, total GC peak area of acetals showed the highest and positive relationship ($r \ge 0.95$) with age, whereas the volatile phenols showed the lowest one ($r \leq 0.05$). Furthermore, from the 103 volatiles tentatively identified to which belong furans, lactones, volatile phenols and acetals, 3-furfural, ethyl 2-furoate, pantolactone, γ -ethoxybutyrolactone, γ -heptalactone, trans-oaklactone, and cis-oak-lactone, p-cresol, trans-dioxane, trans-dioxolane, 1,1-diethoxy-2-methyl-propane, 2,2-diethoxyethanol, 1-(1-ethoxyethoxy)-pentane, 1,1-diethoxypentane, 2-propyl-1,3-dioxolane, and 1,1-diethoxyhexane exhibited the highest and positive correlations (0.71 < r < 0.99) with wine age. Thus, these compounds may be suggested as potential age markers of Malvasia and Bual wines. In a second step, PCA was applied to autoscaled GC peak areas of furans, lactones, volatile phenols, and acetals, in order to extend the study of potential age markers to all noble varieties (Malvasia, Bual, Sercial, and Verdelho) and to the major variety (Tinta Negra) used to produce Madeira wine, which includes different types of wines and ages as described below.



PC1 (32%)

■3Y ●5Y ▲10Y **○**15Y +17Y ◇18Y □19Y △20Y



Figure 6. $PC1 \times PC2$ scores scatter plot (A) and loadings plot profile (B) of GC peak area (attribution of the peak number is shown in Tables 1 and 2).

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Principal Component Analysis. GC peak area of 103 analytical variables (volatile compounds) of the 23 Madeira wines were submitted to a PCA procedure, in order to search for the main sources of variability, to characterize the samples as a function of the detected compounds, and to establish a possible relationship among furans, lactones, volatile phenols and acetals, and wine age. Figure 6A shows the score scatter plot of the two first principal components (which explains 46% of the total variability of the data set), allowing us to organize the Madeira wines by age (as a function of PC1 axis). Figure 6B represents the corresponding loadings plot profile which establishes the relative importance of each volatile for the observed sample distribution of Figure 6A. PC1, which explains 32% of the total variability, allow one to distinguish Madeira wines as a function of their age. PC2, explaining 14% of the total variability, shows the organization of the Madeira wines according to their type. With the exception of Tinta Negra, which may be used to produce all types of wines, all other varieties were used to produce a specific type of wine. Sweet (Malvasia) and medium sweet (Bual) Madeira wines (PC2 positive) are characterized by 5-methyl-2-furfural and diethoxymethane, whereas dry (Sercial) and medium dry (Verdelho) were placed in PC2 negative and are described by 1,1-diethoxyethane and 2,4,5trimethyl-1,3-dioxolane. As can be observed in Figure 6A, the Madeira wines could be organized by their type, as well as grape variety according to PC2. According to PC1, the Madeira wines with 3 and 5Y are projected in PC1 negative, whereas 10Y is near the origin, except for two Madeira wines, which were located near of 5Y, as these wines correspond to the averaged aging period (blended wines). The Madeira wines with 15, 17, 18, 19, and 20Y are placed in PC1 positive. Moreover, the Madeira wines could be organized by their style (vintage or blends) according to PC1. The blend Madeira wine is projected in PC1 negative and is near the origin, whereas the vintage is placed in PC1 positive. Taking into account the loadings plot (Figure 6B), the chemical compounds used as potential age markers were predominantly acetals, namely, diethoxymethane, 1,1-diethoxyethane, 1,1-diethoxy-2-methylpropane, 1-(1-ethoxyethoxy)-pentane, trans-dioxane, and 2-propyl-1,3-dioxolane, and from the other chemical groups studied, 5-methylfurfural (furan) and cis-oak-lactone showed similar contribution to acetals. Finally, it is important to point out that, from these 8 compounds, cis-oak-lactone, transdioxane, 1,1-diethoxy-2-methyl-propane, 1-(1-ethoxyethoxy)pentane, and 2-propyl-1,3-dioxolane were previously (in Establishment of Potential Age Markers) proposed as potential age markers for the Malvasia and Bual varieties. These results suggest that among the chemical groups under study, the acetals are the most important group that could be used as potential age markers of Madeira wines, independently of the variety and the type of wine. Despite the fact that acetals are formed during the fermentation step, its content increases remarkably during the aging process, which may be explained by the oxidative condition that occurred during this step. These oxidative conditions contribute to the increase of aldehyde content, mainly acetaldehyde, and acetals.¹⁰ Moreover, GC × GC-ToFMS offered a very useful approach to identify these chemical groups due to the orthogonal systems that reduce coelution and improve the quality of the selection of volatile compounds under study, contributing to the establishment of new potential Madeira wine age markers.

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